



UNIVERSIDAD DE INVESTIGACIÓN DE TECNOLOGÍA EXPERIMENTAL YACHAY TECH

Escuela de Ciencias Químicas e Ingeniería

**Título: Material selection for oil production processes based
on actual Ecuadorian oil reserves characteristics**

Trabajo de integración curricular presentado
como requisito para la obtención del título de
Petroquímico(a)

Autor:

BASTIDAS GONZALEZ, RICHARD NELSON

Tutor:

PhD. VILORIA VERA DARIO ALFREDO

Co-Tutor:

PhD. MARVIN JOSE RICAURTE

Urcuquí, Marzo 2021

SECRETARÍA GENERAL
(Vicerrectorado Académico/Cancillería)
ESCUELA DE CIENCIAS QUÍMICAS E INGENIERÍA
CARRERA DE PETROQUÍMICA
ACTA DE DEFENSA No. UITEY-CHE-2021-00021-AD

A los 2 días del mes de julio de 2021, a las 16:30 horas, de manera virtual mediante videoconferencia, y ante el Tribunal Calificador, integrado por los docentes:

Presidente Tribunal de Defensa Dr. DIAZ BARRIOS, ANTONIO , Ph.D.
Miembro No Tutor Dr. PALMA CANDO, ALEX URIEL , Ph.D.
Tutor Dr. VILORIA VERA, DARIO ALFREDO , Ph.D.

El(la) señor(ita) estudiante **BASTIDAS GONZALEZ, RICHARD NELSON**, con cédula de identidad No. **0922931894**, de la **ESCUELA DE CIENCIAS QUÍMICAS E INGENIERÍA**, de la Carrera de **PETROQUÍMICA**, aprobada por el Consejo de Educación Superior (CES), mediante Resolución **RPC-SO-39-No.456-2014**, realiza a través de videoconferencia, la sustentación de su trabajo de titulación denominado: **MATERIAL SELECTION FOR OIL PRODUCTION PROCESSES BASED ON ACTUAL ECUADORIAN OIL RESERVES CHARACTERISTICS**, previa a la obtención del título de **PETROQUÍMICO/A**.

El citado trabajo de titulación, fue debidamente aprobado por el(los) docente(s):

Tutor Dr. VILORIA VERA, DARIO ALFREDO , Ph.D.

Y recibió las observaciones de los otros miembros del Tribunal Calificador, las mismas que han sido incorporadas por el(la) estudiante.

Previamente cumplidos los requisitos legales y reglamentarios, el trabajo de titulación fue sustentado por el(la) estudiante y examinado por los miembros del Tribunal Calificador. Escuchada la sustentación del trabajo de titulación a través de videoconferencia, que integró la exposición de el(la) estudiante sobre el contenido de la misma y las preguntas formuladas por los miembros del Tribunal, se califica la sustentación del trabajo de titulación con las siguientes calificaciones:

Tipo	Docente	Calificación
Miembro Tribunal De Defensa	Dr. PALMA CANDO, ALEX URIEL , Ph.D.	9,3
Presidente Tribunal De Defensa	Dr. DIAZ BARRIOS, ANTONIO , Ph.D.	9,0
Tutor	Dr. VILORIA VERA, DARIO ALFREDO , Ph.D.	9,0

Lo que da un promedio de: **9.1 (Nueve punto Uno)**, sobre 10 (diez), equivalente a: **APROBADO**

Para constancia de lo actuado, firman los miembros del Tribunal Calificador, el/la estudiante y el/la secretario ad-hoc.

Certifico que *en cumplimiento del Decreto Ejecutivo 1017 de 16 de marzo de 2020, la defensa de trabajo de titulación (o examen de grado modalidad teórico práctica) se realizó vía virtual, por lo que las firmas de los miembros del Tribunal de Defensa de Grado, constan en forma digital.*

BASTIDAS GONZALEZ, RICHARD NELSON
Estudiante

Dr. DIAZ BARRIOS, ANTONIO , Ph.D.
Presidente Tribunal de Defensa

ANTONIO DIAZ BARRIOS
 Firmado digitalmente por ANTONIO DIAZ BARRIOS
 Fecha: 2021.07.13 22:06:42 -05'00'

Dr. VILORIA VERA, DARIO ALFREDO , Ph.D.
Tutor

DARIO ALFREDO VILORIA VERA
 Firmado digitalmente por DARIO ALFREDO VILORIA VERA
 Fecha: 2021.07.13 19:39:47 -05'00'



Firmado electrónicamente por:
ALEX URIEL
PALMA CANDO

Dr. PALMA CANDO, ALEX URIEL , Ph.D.

Miembro No Tutor

CARLA SOFIA

YASELGA NARANJO

Digitally signed by CARLA SOFIA YASELGA NARANJO
Date: 2021.07.07 17:26:35 -0500

YASELGA NARANJO, CARLA

Secretario Ad-hoc

AUTORÍA

Yo, **RICHARD NELSON BASTIDAS GONZALEZ**, con cédula de identidad 0922931894, declaro que las ideas, juicios, valoraciones, interpretaciones, consultas bibliográficas, definiciones y conceptualizaciones expuestas en el presente trabajo; así cómo, los procedimientos y las herramientas utilizadas en la investigación, son de absoluta responsabilidad de el/la autor(a) del trabajo de integración curricular. Así mismo, me acojo a los reglamentos internos de la Universidad de Investigación de Tecnología Experimental Yachay.

Urcuquí, Marzo 2021.



Richard Nelson Bastidas González
C.I.: 0922931894

AUTORIZACIÓN DE PUBLICACIÓN

Yo, **RICHARD NELSON BASTIDAS GONZALEZ**, con cédula de identidad 0922931894, cedo a la Universidad de Tecnología Experimental Yachay, los derechos de publicación de la presente obra, sin que deba haber un reconocimiento económico por este concepto. Declaro además que el texto del presente trabajo de titulación no podrá ser cedido a ninguna empresa editorial para su publicación u otros fines, sin contar previamente con la autorización escrita de la Universidad.

Asimismo, autorizo a la Universidad que realice la digitalización y publicación de este trabajo de integración curricular en el repositorio virtual, de conformidad a lo dispuesto en el Art. 144 de la Ley Orgánica de Educación Superior.

Urcuquí, Marzo 2021.



Richard Nelson Bastidas González
C.I.: 0922931894

Dedication

This work is dedicated to my family, specially to my parents!

Acknowledgements

I would like to say thanks to my parents because they are my support for every achievement that I have achieved. They are my motivation and my inspiration.

I would like to say thanks to all of my family, because they always believe in me.

I would like to say thanks to my teachers, because each one has given me a teaching, and always stay to clear all my doubts and allow me to complete a correct professional formation.

Resumen

En la actualidad, los yacimientos de crudo en Ecuador contienen crudo más pesado que los que se procesaban cuando se construyeron las facilidades de producción, tales como campos para la exploración del petróleo, oleoductos, refinerías, etc. La declinación natural de yacimientos conlleva a un aumento de la producción de agua, lo que impacta considerablemente a la confiabilidad y continuidad operacional; por lo tanto, los problemas de corrosión en todo el proceso involucrado desde el campo petrolífero hasta el producto final han aumentado. Una adecuada selección de materiales, así como el uso de tratamientos químicos deben ser considerados desde las etapas de diseño; considerando como principal campo petrolero el campo Sacha, las características de su crudo y las condiciones de operación a las que será sometido el material. Para este estudio fue necesario analizar los posibles problemas de corrosión que se pueden presentar. Para la determinación de la formación de escamas se utilizó el modelo Langelier y el modelo Stiff-Davis, considerando las condiciones de boca de pozo, en cuyo caso se descartaron con base en los valores obtenidos de -0,04 y 0,11 respectivamente. Para la determinación de la corrosión por CO₂ se utilizaron los modelos De waar-Milliams y De waard-Lotz, en cuyo caso se desarrolló y utilizó un perfil de presión y temperatura para obtener los valores estimados para cada punto del pozo de producción. Según De waar-Milliams, las velocidades de corrosión varían de 1,25 mm / año a 39,73 mm / año y según De waard-Lotz pueden variar de 4,40 mm / año a 58,62 mm / año; en ambos casos los valores obtenidos representan un problema operacional inminente, por lo que es necesario considerar el uso de una tubería de acero de altas prestaciones. Según la Tabla de Selección de Materiales propuesta por Vera de DNV.GL con una presión parcial de CO₂ del orden de 10² y una presión parcial de H₂S insignificante, el material que se recomienda utilizar es el 13-Cromo.

Palabras claves: CRA's, cloruros, H₂S, datos PVT, métodos de producción, integración subsuelo-superficie, carga frontal.

Abstract

Nowadays, Ecuador's crude oil reservoirs contain heavier crude oil than those processed when production and processing facilities, such as oil fields exploration, pipelines, and refineries. The natural decline of reservoirs increases the water production, which significantly impacted reliability and operational continuity; therefore, the corrosion problems in all of the oilfield processes to the final product have increased. An adequate material selection and the chemical treatment usage have to be considered from the design steps, considering Sacha oilfield as the principal oilfield in Ecuador and its crude oil characteristics and its operational conditions to which the material will be subjected. For this study was necessary to analyze the possible corrosion problems that could present. Langelier model and Stiff-Davis model were used for scale formation determine, considering wellhead conditions, in which case were discarded based on the values obtained of -0.04 and 0,11 respectively. De waar-Milliams and De waard-Lotz models were used for CO₂ corrosion determine, in which case a pressure and temperature profile were developed and used in order to obtain the estimated values for each point of the production well. According to De waar-Milliams the corrosion rates varies from 1,25 mm/yr to 39,73 mm/yr and to De waard-Lotz it can varies from 4,40 mm/yr to 58,62 mm/yr; in both cases the values obtained represent an imminent operational problem, for this reason the use of a high-performance steel pipe is necessary to be considered. According to Material Selection Table propoused by Vera from DNV.GL with a CO₂ partial pressure in the order of 10² and a negligible H₂S partial pressure the material that is recommended to be used is 13-Chrome.

Keywords: CRA's, chlorides, H₂S, CO₂, PVT data, production methods, subsurface-surface integration, front end loading.

TABLA DE CONTENIDO

CHAPTER 1: INTRODUCTION	1
1.1 Problem statement	1
1.2 Objectives	1
1.2.1 General Objective	1
1.2.2 Specific Objectives	1
CHAPTER 2: BACKGROUND	3
2.1 Corrosion in the hydrocarbon industry	3
2.1.1 Principal Corrosion problems in hydrocarbon industry	4
2.2 CRA's	6
CHAPTER 3: SACHA'S OILFIELD	7
3.1 Sacha's Oilfield	7
3.2 Geological Structure of Sacha's Oilfield	11
3.3 Sacha's Oil Characteristics	12
3.4 PVT data	13
3.5 Formation Water Composition	13
CHAPTER 4: METHODOLOGY	15
4.1 Scale Formation Prediction	16
4.1.1 Langelier Saturation Index (LSI)	17
4.1.2 Stiff-Davis saturation index (ISSD)	19
4.1.3 Oddo-Tomson saturation index	21

4.2 CO ₂ Corrosion Prediction.....	24
4.2.1 De waard-Milliams Method.....	24
4.2.2 De waard-Lotz model	24
4.3 Material Selection Diagram	25
CHAPTER 5: RESULTS AND DISCUSSION.....	27
5.1 Hydrocarbon industry in the world	27
5.2 Americas oil industry situation	29
5.3 Ecuadorian hydrocarbon industry	31
5.3.1 Principal processes for production in Ecuador	31
5.3.2 Actual situation of Ecuadorian Oilfields	32
5.4 Scale Formation Results.....	35
5.4.1 Langelier results	35
5.4.2 Stiff-Davis Results.....	37
5.5 CO ₂ Corrosion Prediction Results.....	38
5.5.1 De waard-Milliams Results	38
5.5.2 De waard-Lotz Results	43
5.6 Discussion	45
CONCLUSION AND RECOMMENDATIONS	48
REFERENCES	49
ANNEX.....	54

List of Figures

<i>Fig. 1. CO₂ Corrosion Mechanism (Viloria, 2014)</i>	4
<i>Fig. 2. H₂S corrosion scheme (Himipex Oil Ltd, 2015)</i>	5
<i>Fig. 3. Sacha's Oilfield Location</i>	7
<i>Fig. 4. Total proved reserves per asset</i>	8
<i>Fig. 5. Annual production and income per field</i>	9
<i>Fig. 6. Actual condition of Sacha field reserves</i>	10
<i>Fig. 7. Stratigraphic Column of the eastern basin</i>	11
<i>Fig. 8. Methodology Diagram</i>	16
<i>Fig. 9. Correlation for K values determine (Patton, 1995)</i>	20
<i>Fig. 10. Material selection diagram (Vera, 2016)</i>	26
<i>Fig. 11. Relation reserve-production of oil countries</i>	27
<i>Fig. 12. Ecuadorian Oilfields(Hydrocarbon Secretariat, 2016)</i>	33
<i>Fig. 13. Graphic temperature vs LSI</i>	36
<i>Fig. 14. Corrosion rate profile obtained by (Eq. 15)</i>	41
<i>Fig. 15. Corrosion rate profile obtained by (Eq. 16)</i>	42
<i>Fig. 16. Corrosion rate profile obtained by De waard-Lotz model</i>	44

List of Table

<i>Table. 1. Probability of corrosion based on CO₂ partial pressure (Castellan, 1987)</i>	5
<i>Table. 2. Sacha's crude oil characteristics (Fernandez&Gaibor, 2009)</i>	13
<i>Table. 3. PVT Data Sacha Oilfield (Romero.J.&Gomez.F.,2010)</i>	13
<i>Table. 4. Formation water composition (Carrillo,2017)</i>	14
<i>Table. 5. Reserve, Annual Production and rate of Principal oil countries in America</i>	29
<i>Table. 6. Oil Consumption of principal oil countries in America</i>	30
<i>Table. 7. Probable and Possible reserves principal oilfields</i>	34
<i>Table. 8. Langelier Method results for Sacha's formation water</i>	35
<i>Table. 9. Results of ions' molar concentrations of the formation water</i>	37
<i>Table. 10. Result of Molar Ionic Strength</i>	37
<i>Table. 11. K value obtained by the correlation diagram</i>	37
<i>Table. 12. Stiff-Davis saturation index obtained</i>	38
<i>Table. 13. Partial pressure of CO₂</i>	39
<i>Table. 14. De waard-Milliam results of corrosion rate profile</i>	40
<i>Table. 15. Liquid flow rate and diameter for a typical Sacha's oilfield</i>	43
<i>Table. 16. Corrosion rate obtained by De waard-Lotz model</i>	44

ABBREVIATIONS

GE	Specific Gravity
BSW	Basic Sediments and Water
T	Temperature
GOR	Gas to Oil Ratio
Boi	Volumetric Factor of the Oilfield
TDS	Total Dissolved Solids
pHs	Saturation pH
pCa	Negative Logarithm of Calcium Concentration
pHCO ₃	Negative Logarithm of Bicarbonate Concentration
p <i>K_{sp}</i>	Negative Logarithm of Solubility Product Constant
pK ₂	Negative Logarithm of the Bicarbonate Ionization Constant
K	Constant of Salinity
LSI	Langelier Saturation Index
ISSD	Stiff-Davis Saturation Index
pAlk	Negative Logarithm of Alkalinity
Si	Ionic Strength
Zi	Ion Valence
Ci	Ion Concentration
<i>f_g^{CO2}</i>	Fugacity of CO ₂
<i>y_g^{CO2}</i>	The CO ₂ Molar Fraction in the Gas Phase
<i>y_t^{CO2}</i>	The CO ₂ Molar Fraction in the Surface
<i>C_{aq}^{CO2}</i>	The CO ₂ Concentration in Aqueous Phase

CRt	Corrosion Rate (De waard-Milliams Model)
Vcr	Corrosion Rate (De waard-Lotz Model)
Vr	Independent Flow Contribution
Vm	Dependent Flow Contribution
U	Liquid Flow Rate
Dh	Hydraulic Diameter
PCO ₂	Partial Pressure of CO ₂
MMscf	Million Standard Cubic Feet
BWPD	Barrels of Water per Day
BOPD	Barrels of Oil per Day
BFPD	Barrels of Fluid per Day

CHAPTER 1: INTRODUCTION

1.1 PROBLEM STATEMENT

The Ecuadorian recovery factor has changed over time because the principal methods of production have to be changed based on the production decline. Initially, the production method was by hydraulic push. Nowadays, recovery methods such as water reinjection have to be used specially in the principal oilfields such as Sacha located at Ecuadorian east in the Orellana province in order to maintain its production. The change of the recovery factor and an inadequate material selection can increase the corrosion problems presented in the different pipelines, process equipment or deposits, producing the deterioration of the facilities and even catastrophic events. Nowadays, Sacha oilfield represents the country's highest income in terms of oilfields, and according to its reserves, it will continue representing. Its actual reserve has changed across time as everywhere. Therefore, for this reason, an analysis of its current conditions is necessary to determine the correct material selection for future projects.

1.2 OBJECTIVES

1.2.1 GENERAL OBJECTIVE

- ❖ Study and identify the impact of the recovery factor on the corrosion behavior in Sacha oilfield facilities in order to determine the suitable material selection.

1.2.2 SPECIFIC OBJECTIVES

- ❖ Study the hydrocarbon industry's present situation, oil reserves globally, and Ecuadorian oilfields' current and future situation.
- ❖ Study the Sacha oilfield impact on the country's economy, its chemical composition of crude oil and the operational conditions.

- ❖ Study and analyze the possible corrosion problems presented in Sacha oilfield, by using the principal corrosion identification models.
- ❖ Analyze and interpret the results obtained to propose the correct material to be used in future Sacha's wells based on the results obtained.

CHAPTER 2: BACKGROUND

2.1 CORROSION IN THE HYDROCARBON INDUSTRY

Corrosion is one of the principal problems in the hydrocarbon industry from the beginning, representing one of the major losses of money and generating possible catastrophic disasters.

According to NACE's basic manual, "Corrosion is the deterioration of a material, especially metals resulting from its contact with the environment" (NACE, 2014).

The fight with the corrosion problems in the oil industry can present in different sceneries and by different factors that can be external or internal. The external factor is the environment, which begins in the oilfield and finishes in the refinery location. The internal factors depend on the crude oil's chemical characteristics, especially in chlorides content, CO₂ or H₂S content.

According to the Chemical review "Corrosion – The longest war" , the corrosion maintenance for BP represents from 25 to 30% of their operational cost. This is an important data to consider because engineers need to have present that it can increase depends on the environmental conditions, such as it is the Ecuadorian case.

According to the actual Petroecuador EP annual budget in 2015, the money earmarked for operation and maintenance service is 577.285,738 USD as it is shown in the Annex A.

Corrosion is extremely important for the hydrocarbon industry, especially if the economic and financial aspects are considered because it represents inspections and maintenance costs or even if the environmental impact is considered. There are two types of associated costs, one that is directly related to corrosion and the other indirect related to it. There are costs such as maintenance, loss of production, product contamination, cost of overdesign or environmental penalties for the first type. The indirect ones are security, the collapse of nearby structures, leaks or indirect product contamination.

Corrosion cannot be avoided, but it can be controlled, and its impact can be reduced. There are different ways, but the principles are using good chemical protection and the correct material selection based on the environmental conditions, the operational conditions and the crude oil characteristics to control the main corrosion problems presented.

2.1.1 PRINCIPAL CORROSION PROBLEMS IN HYDROCARBON INDUSTRY

2.1.1.1 CO₂ CORROSION

CO₂ corrosion is one of the most common corrosion problems in the hydrocarbon industry, and it could be known as sweet corrosion. It can be produced by different possible sources such as formation fluids, aeration, make-up water, or bacterial activity. In most cases, it is caused when the CO₂ has contact with the production or transportation water, and the principal affected areas are the internal well section or the transport lines.

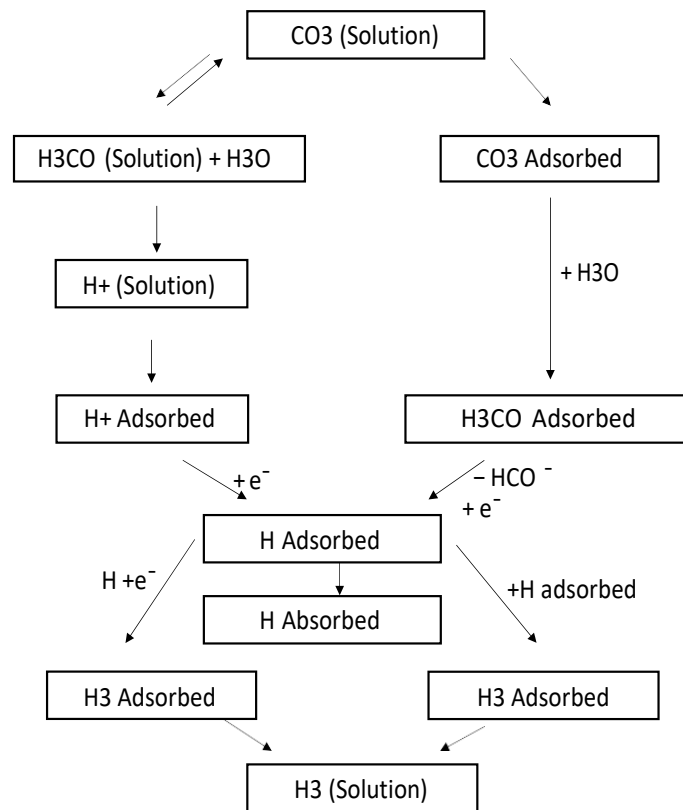


FIG. 1. CO₂ CORROSION MECHANISM (VILORIA, 2014)

The reaction between CO₂ and water gives carbonic acid, which is a weak acid. Then, the dissociation reaction occurs rapidly, especially at high temperatures, and finally, the reaction with iron occurs to produce ferrous carbonate. The formation of a ferrous carbonate scale on the surface generates iron passivation, which gives way to corrosion.

According to the NACE standards, the partial pressure of CO₂ is an essential parameter to determine if corrosion by CO₂ will be present, as shown in Table. 1.

TABLE. 1. PROBABILITY OF CORROSION BASED ON CO₂ PARTIAL PRESSURE (CASTELLAN, 1987)

Partial pressure of CO ₂ (bar)	Probability of corrosion
< 0,5	Less probable
0,5 < P _{CO2} < 2	Possible
> 2	Assured

This kind of corrosion can be controlled with the use of inhibitors or corrosion-resistant alloys, commonly known as CRA's.

2.1.1.2 H₂S CORROSION

H₂S corrosion is one of the most common corrosion problems in the hydrocarbon industry, and it could be known as acid corrosion. It can be produced by different possible sources such as formation fluids, the degradation of drilling muds, or bacterial activity.

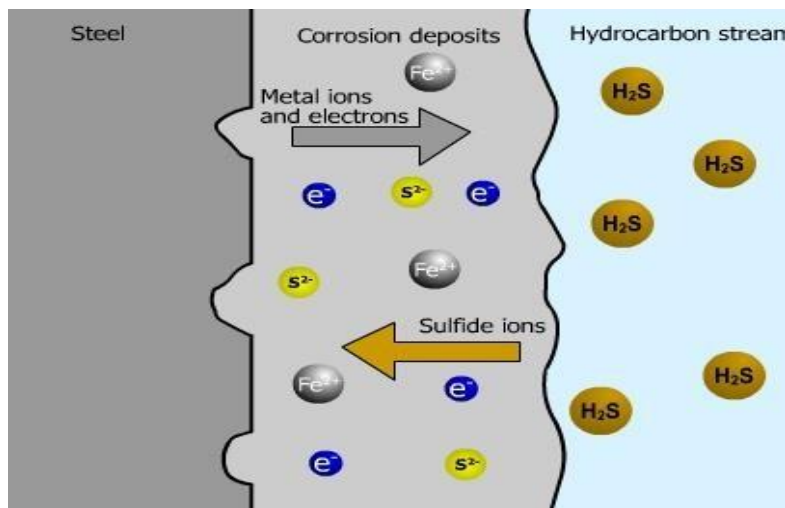


FIG. 2. H₂S CORROSION SCHEME (HIMPEX OIL LTD, 2015)

The hydrogen sulfide can affect any part of the production system at the moment of contact with water, and it is necessary to have a concentration of 5 ppm in the formation water to consider the production fluid as acid.

When the hydrogen sulfide reacts with different iron oxides to form iron sulfides, those iron sulfides increase the corrosion velocity. When iron sulfides precipitate, micro galvanic cells can form, which generates localized corrosion such as pitting or cracks. This corrosion problem can be controlled by using specific materials, which can form an organic film to sequester H₂S. Other options can be metals resistant to sulfur stress corrosion, nitrate treatments or biocides.

2.2 CRA'S

"CRA is an alloy intended to be resistant to general and localized corrosion or environmental cracking in environments that are corrosive to carbon and low-alloy steels" (ISO 13680). Corrosion-resistant alloys' principal objective is to control corrosion in the well, and they are chosen because chemical inhibition is not practical. The normative requirements are the API5CRA and NACE MR0175. The principal CRA's that are available for the hydrocarbon industry are Martensitic Stainless Steels, Duplex Stainless Steel and Nickel Alloys. In order to determine if it is necessary to use a corrosion-resistant alloy is important to analyze the following factors: H₂S partial pressure, CO₂ partial pressure, chloride content, produced water rate, bicarbonate content, free sulfur content and the operational parameters because all of these components determine the corrosion impact.

One of the principal materials used in the hydrocarbon industry is the high-performance steels to supply the necessity and the high demands concerning productivity, reliability and security. High-performance steels increase the structure's lifetime and increase production time, which is essential in the hydrocarbon industry and reduces maintenance costs.

CHAPTER 3: SACHA'S OILFIELD

3.1 SACHA'S OILFIELD

The Sacha Oilfield known as the block 60, which is located in the Orellana province at the northeast of the Ecuadorian Amazonia. It is delimited by Eno-Ron oilfield at the north, Culebre-Yulebra oilfield in the south, Shushufindi at the east and Paraiso at the west as it is shown in the Fig. 3.

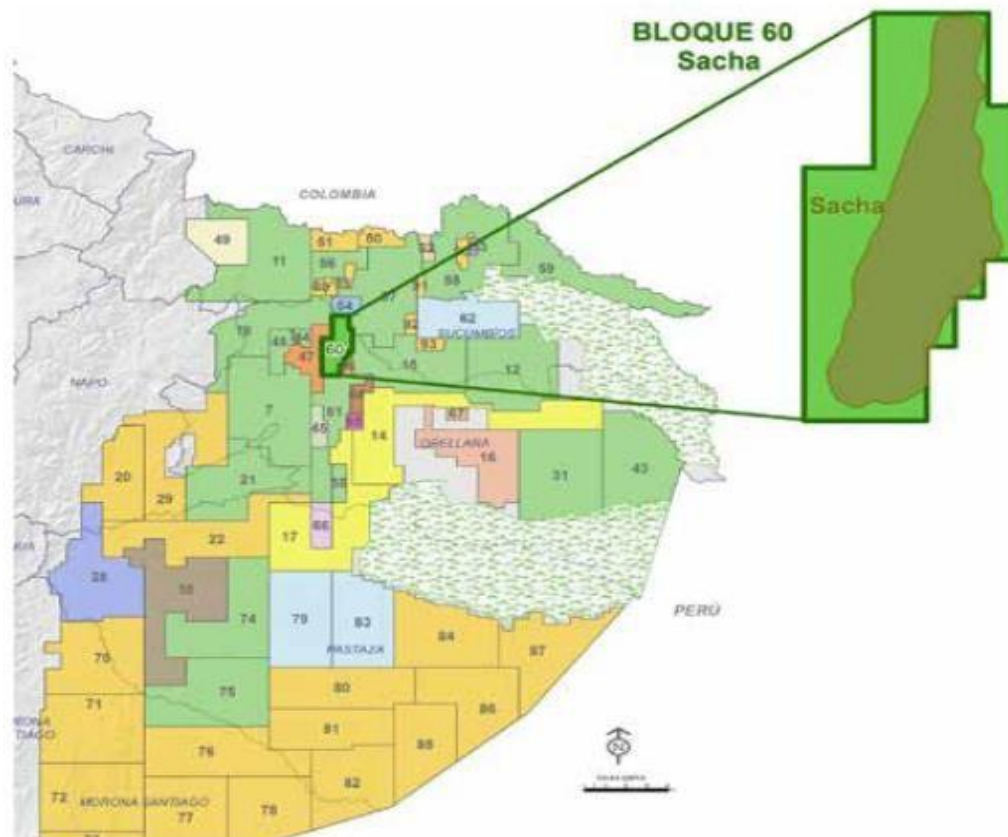


FIG. 3. SACHA'S OILFIELD LOCATION

(ANNUAL REPORT OF THE HYDROCARBON POTENTIAL OF ECUADOR 2018)

Directly cited from: <https://www.recursosyenergia.gob.ec/wp-content/uploads/2019/11/Informe-Anual-del-Potencial-Hidrocarburi%CC%81fero-del-Ecuador-2018.pdf>

This oilfield has been operated by Petroecuador EP since 2021. However, Sacha oilfield was discovered by the Texaco-Gulf consortium in 1969 and immediately passed to be one of the most important oilfields in Ecuador. Sacha currently produces 71,400 barrels per day. This is why it is considered one of the "jewels of the corona" of the Ecuadorian oil sector. Sacha is referenced as one of the most productive oilfields in the Ecuadorian Hydrocarbon industry.

According to the Annual Report of The Hydrocarbon Potential of Ecuador 2018, the total oil reserve in Ecuador is 2,695 million barrels divided into 1,632 million barrels of proved reserves, 314 million barrels of probable reserves and 749 million barrels of possible reserves. The 88.33% of the total oil reserve in Ecuador is managed by the public company Petroamazonas EP. This percentage comprises 1,390 million barrels of proved reserves, 259 million barrels of probable reserves and 731 million barrels of possible reserves. Based on the data shown in Fig. 4 and Fig. 5 Sacha's Oilfield play an important role in the Ecuadorian hydrocarbon industry.

ECUADORIAN TOTAL PROVED RESERVES

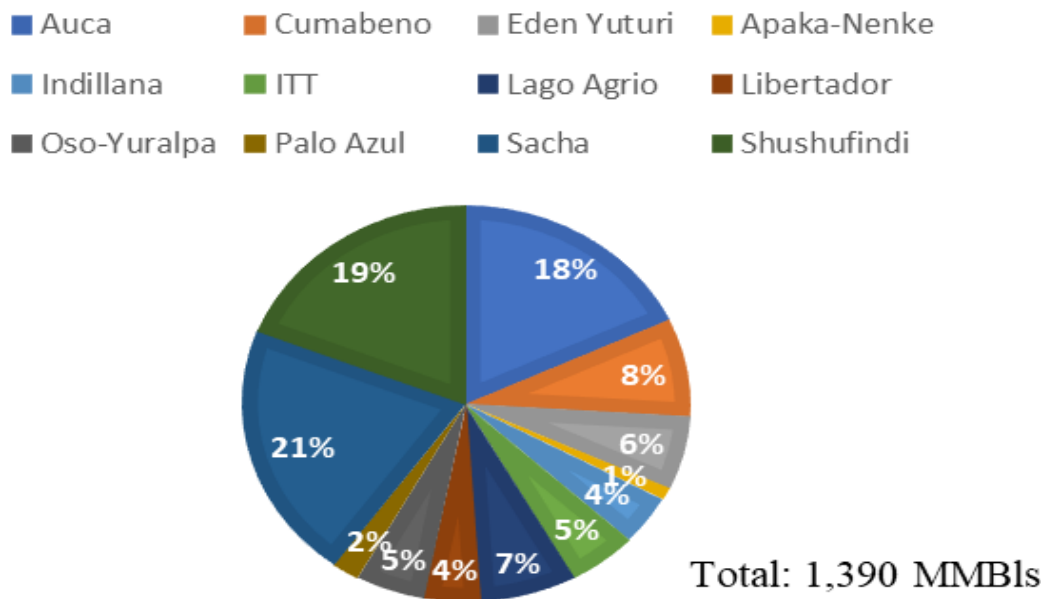
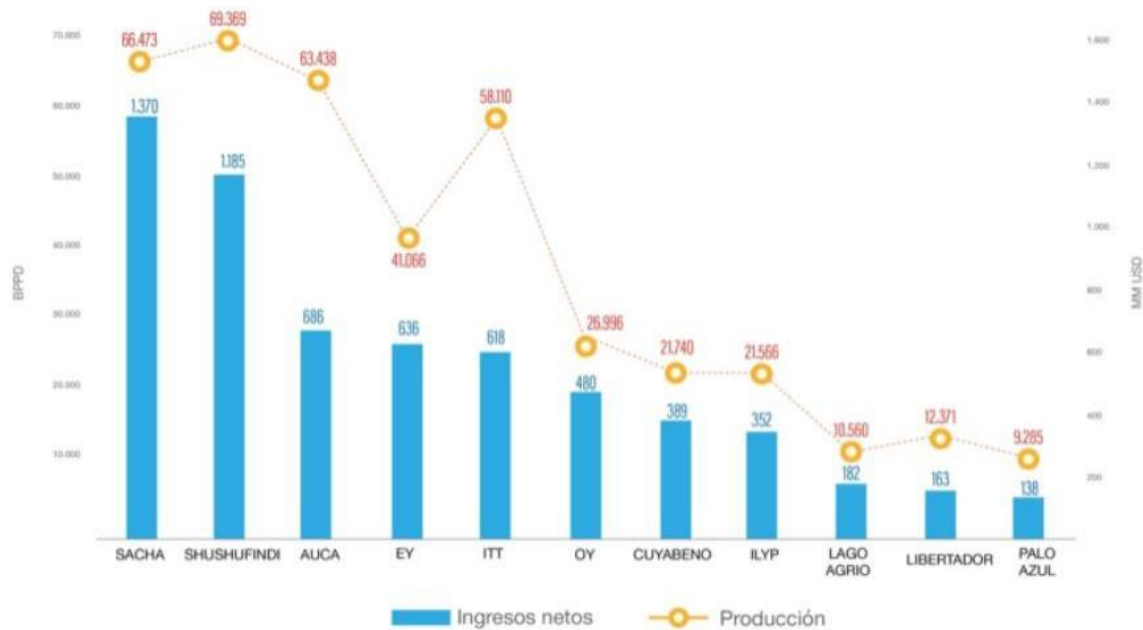


FIG. 4. TOTAL PROVED RESERVES PER ASSET

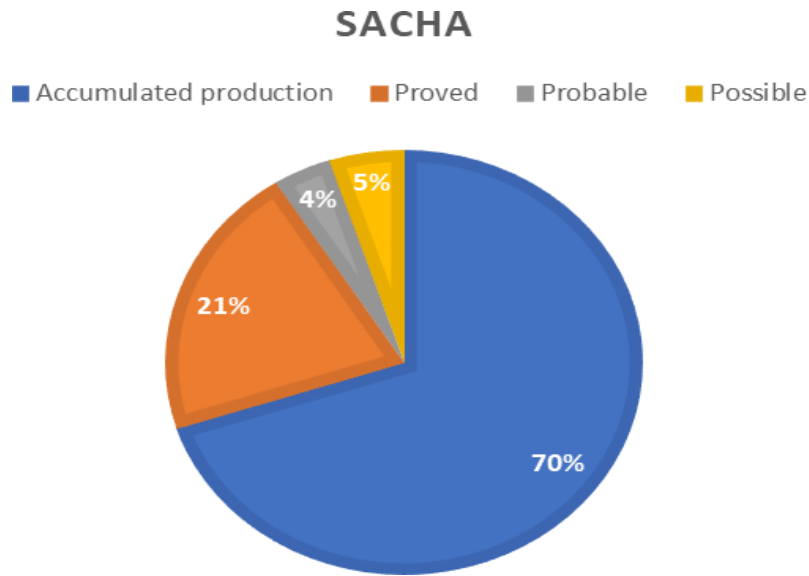
(ANNUAL REPORT OF THE HYDROCARBON POTENTIAL OF ECUADOR, 2018)



*FIG. 5. ANNUAL PRODUCTION AND INCOME PER FIELD
(PETROAMAZONAS, 2018)*

According to Fig. 4 and Fig. 5 Sacha has the biggest proved reserve in Ecuador with 291,9 million barrels with a production of 66.473 barrels per day and an annual income of 1370 million dollars in 2018. These numbers make Sacha the oilfield with the highest income in Ecuador even it does not have the highest daily production. This highest income can be credited to its crude oil characteristics, which are better than in the other oilfields.

Based on the data given by the annual report of the hydrocarbon potential of Ecuador, Sacha has 419,06 million barrels between proved, probable and possible reserve, which represent the 15.55% of the total oil reserve in Ecuador. Therefore, the study of Sacha oilfield is necessary to know its current condition and the best way to optimize and take advantage of the field resources and preserving its facilities.



*FIG. 6. ACTUAL CONDITION OF SACHA FIELD RESERVES
(ANNUAL REPORT OF THE HYDROCARBON POTENTIAL OF ECUADOR, 2018)*

According to Fig. 6 Sacha field has 125.72 million barrels approximately to explode, which with a constant production represents more than 5 years of life time.

Nowadays, water reinjection method is used in the different wells in Sacha field in order to increase the oil recovery. This increase at the same time the probability of corrosion based on the formation water characteristics and the environment; because open system used for reinjection water generate corrosion by microorganism, corrosion by CO₂ or by the chlorides present in the formation water. These corrosion problems lead to the abandonment of a well such as SAC-100 (Landazuri, 2015).

The Sacha oilfield conditions, the projection for the future and corrosion problems presented motivate the engineers to analyze the chemical control used and the metallurgy selection, because the downhole conditions, the crude oil characteristics and the environmental conditions are affecting and increasing the corrosion problems presentes in the processes involved in the hydrocarbon industry from the downhole to the final consumer.

3.2 GEOLOGICAL STRUCTURE OF SACHA'S OILFIELD

Sacha oilfield is formed by an asymmetric low relief anticline failed to the west; the field is cut by fault that originates in the basement and which probably controlled the sediments deposition. The oilfields in Ecuador in the eastern basin are related to the mesozoic era between the lower, medium and upper cretaceous that correspond to the Hollin, Napo and Basal Tna formation as it is shown in the stratigraphic column in the Fig. 7 (Romero and Gomez, 2010).

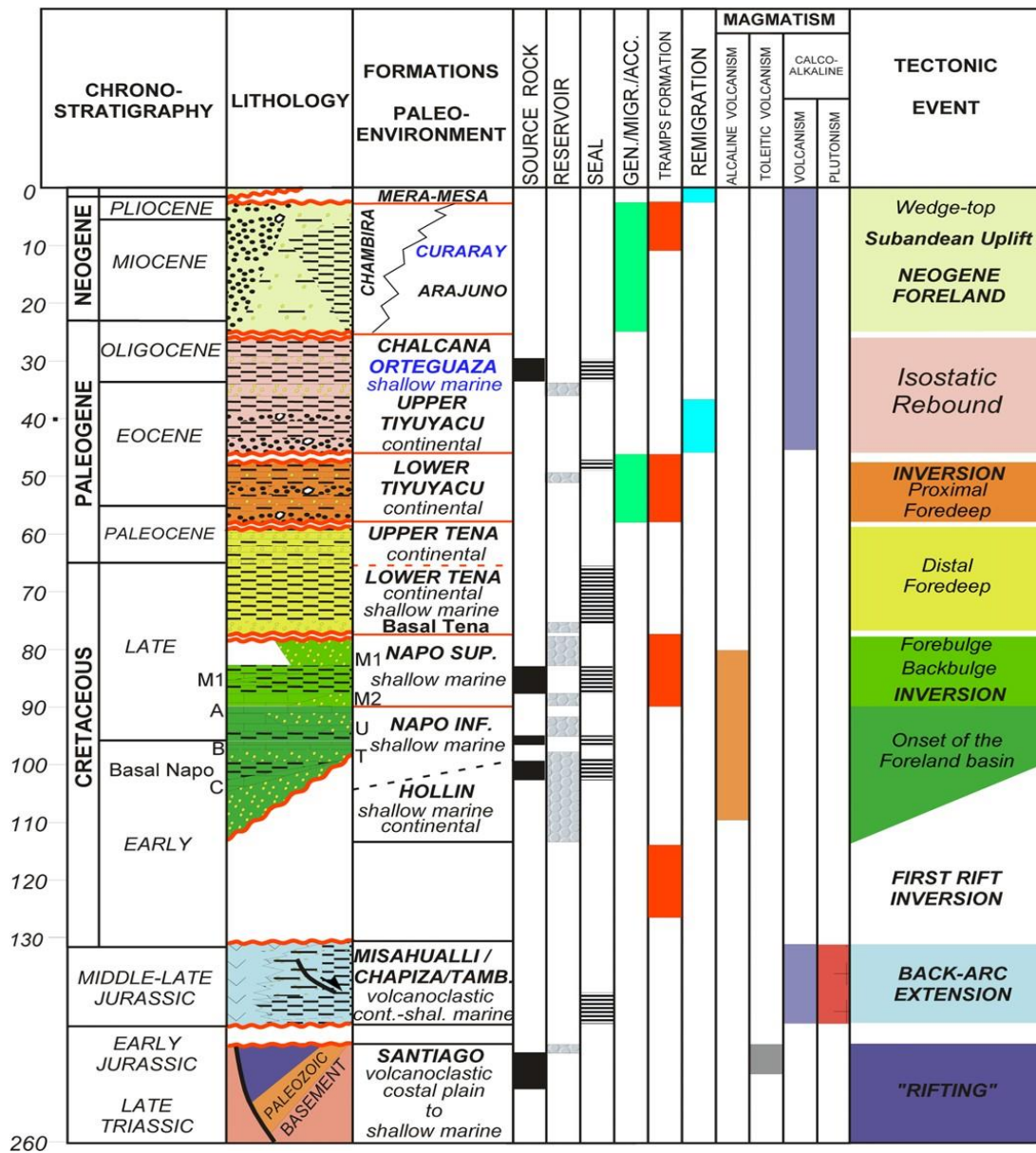


FIG. 7. STRATIGRAPHIC COLUMN OF THE EASTERN BASIN

(BABY ET. AL, 2013)

The four main strata of the stratigraphic column are Hollin, Napo T, Napo U and Tena; their importance is based on their content of crude oil and gas.

The Basal Tena lithology is composed by sandstone with a porosity of 19 percent, no average permeability, recovery factor from 20-30 percent, and no gas-oil relation. Napo is divided into two sections Napo U, and Napo T. Napo U is composed basically of sandstone and shale with a porosity of 17 percent, an average permeability of 210 MD, a recovery factor of 39 percent and a gas and oil relation of 220, which means that is suitable for gas production. Napo T is composed of sandstone, shale and limestone with a porosity of 14 percent, an average permeability of 325, a recovery factor of 33 percent and a gas-oil relation of 325. Finally, the most important is Hollin, which is composed of sandstone. It has an average porosity of 16 percent, an average permeability of 690, a recovery factor of 33 percent and a gas-oil relation of 14; this means that the principal strata of interest is Hollin, because it is where the crude oil comes from for production (Romero and Gomez, 2010).

The percentage of water and the average salinity percentage are important aspects of the oilfield when a production project is running. The Basal Tena has an average salinity from 24000 to 36000 ppm Cl⁻ and a 31 percent of water saturation. The Napo U has an average salinity from 25000 to 45000 ppm Cl⁻ and 25 percent of water saturation. Napo T has an average salinity from 20000 to 25000 ppm Cl⁻ and 14.8 percent of water saturation. Finally, Hollin has an average salinity from 500 to 3500 ppm Cl⁻ and 35 percent of water saturation (Romero and Gomez, 2010).

3.3 SACHA'S OIL CHARACTERISTICS

The crude oil characteristics are absolutely important to determine how to treat the crude oil and the production method that will be used. The principal characteristics are the API gravity, the specific gravity, the viscosity and the water&sediments percentage, as shown in the Table. 2. In Sacha Oilfield's case, it is divided in two principal zones are South Sacha and North Sacha.

TABLE. 2. SACHA'S CRUDE OIL CHARACTERISTICS (FERNANDEZ AND GAIBOR, 2009)

Characteristics	South Sacha	North Sacha
API°	26,9	25,2
GE	0,895	0,903
Viscosity (cp at 80°F)	50,6	58,74
Viscosity (cp at 120°F)	13,54	21,63
BSW (%)	35,38	56,61

3.4 PVT DATA

The pressure, volume and temperature are important parameters to be considered when oilfields are producing because they are the initial operation condition considered when the production method will be selected. In Sacha oilfield, the PVT data is divided based on the different formations shown in Table. 3.

TABLE. 3. PVT DATA SACHA OILFIELD (ROMERO AND GOMEZ, 2010)

Parameter	Reservoir				
	BT	U	T	Hs	Hi
Temperature (°F)	181	219	221	225	225
Initial pressure (psi)	3587	4054	4146	4450	4450
Bubble pressure (psi)	870	1170	1310	550	80
GOR (SCF/B)	150	284	389	93	93
API°	24,1	22,8	30,3	27,3	29,7
Boi (BY/BN)	1,117	1,2302	1,3726	1,1334	1,1625

3.5 FORMATION WATER COMPOSITION

The formation water is the one that is coming from the reservoir, from lateral formations such as aquifers or the reinjection process known as secondary recovery. It is a big percentage of the production fluid, especially in the actual crude oil characteristics. The composition of the

formation water can be different in each producing well based on the different kinds of geological formations, the treatment of the crude oil, the production method and the chemicals used. In general, this water has many minerals in the saturation limit, and when they go beyond, they can generate scale.

TABLE. 4. FORMATION WATER COMPOSITION (CARRILLO,2017)

Property	Unids	Value
Total Hardness	ppm	2200
Calcium Hardness	ppm	1800
Magnesium Hardness	ppm	400
Alkalinity	ppm	350
Na ⁺	ppm	5759
Calcium ion	ppm	720
Magnesium ion	ppm	96
HCO_3^-	ppm	434
Iron ion	ppm	4,5
Sulfate ion	ppm	3
Cl ⁻	ppm	10200
pH		6,33
TDS	ppm	17223
Temperature	°F	127
CO3 in brine	ppm	330
H3S in brine	ppm	1,5
Disolved oxygen	ppb	10
Haze	NTU	23,4
SST	ppm	29
Oil in brine	ppm	22,3
Residual Fosfate	ppm	27

CHAPTER 4: METHODOLOGY

To make the correct material selection for a production well in an specific oilfield in this case Sacha's Oilfield was necessary to follow the steps described in Fig. 8. First, it was necessary to study and determine three important factors, which are the actual industry situation, the principal corrosion problems presented in the hydrocarbon industry and determine the oilfield to be studied (case study). Second, it was necessary to find all the available information about the operational conditions and oilfield characteristics in order to analyze them and identify the possible corrosion problems that could present. Third, based on the possible corrosion problems and on the available oilfield data and characteristics, the principal correlation models used to determine corrosion in oil production were considered and selected. In the case of scale formation prediction Langelier saturation index was used and the Stiff-Davis saturation index was used to reconfirm and compare with the LSI obtained. In the case of CO₂ corrosion prediction De waard-Milliams and De waard-Lotz models were used in order to obtain the corrosion rate and know the CO₂ impact in the pipeline used. Finally, with both results the material selection diagram shown in Fig. 10 was used to determine the adequate material for Sacha's Oilfield conditions and stream, and contrasted with the actual material used in Sacha's oilfields which is based on API 5L GRB.

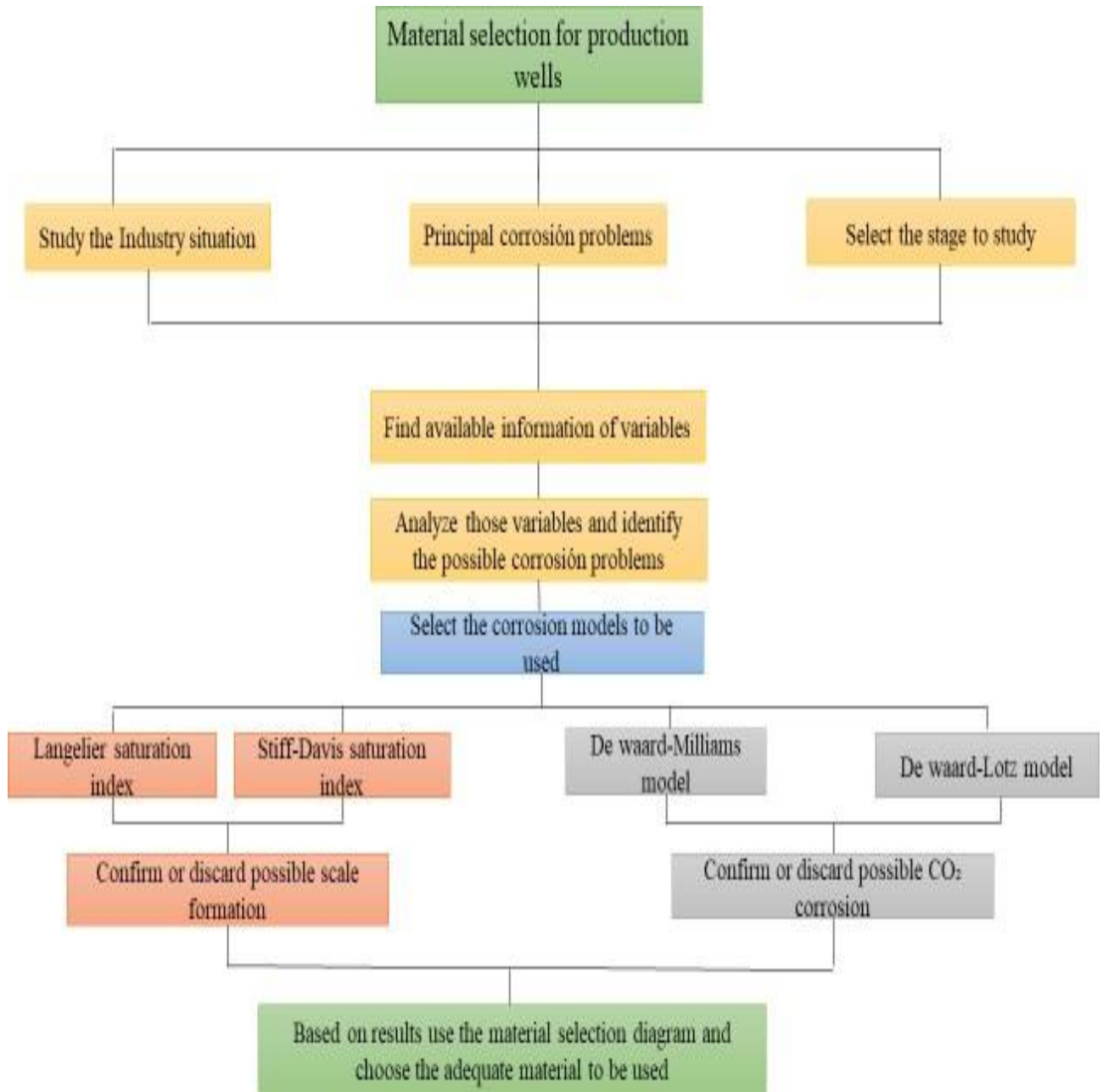


FIG. 8. METHODOLOGY DIAGRAM

4.1 SCALE FORMATION PREDICTION

Different models can predict the scale formation to determine the fouling potential in the different oil fields. The principal methodology used for Scale formation prediction is

based on the saturation conditions evaluated by a saturation index or saturation grade. The saturation index allows engineers to calculate and determine the scale formation capacity based on the formation water's physical-chemical characteristics. Nowadays, the saturation index can be determined by different mathematical models that evaluate the effects caused by the variation of pH, temperature, pressure in the formation water to generate solid deposits. The principal methods used are the Langelier Saturation index, Stiff-Davis index and Odde-Tomson index.

4.1.1 LANGELIER SATURATION INDEX (LSI)

The Langelier saturation index gives the grade of Saturation of the formation water concerning calcium carbon formation. The formation water is part of the production fluid, which is why it is absolutely important to consider it to discard this incrustation in all production lines. Langelier can be interpreted as the change in pH required to reach the water equilibrium. In other words, is the difference between the current pH and the saturation pH necessary to avoid changes in the composition.

This saturation index model's main restriction is that the Langelier Saturation model is valid with constant temperature for moderate salinity and systems with $\text{pH} < 9$. This model takes the K_2 (second dissociation) and the K_{sp} (constant of the product solubility) as values that only depend on the ion concentration, which means that they do not depend on the temperature, pressure or the ionic force. Another important point to consider is that the pH values used are taken at the wellhead not at the downhole, which in the oil production case is the most critical point considering that it has a different and higher temperature. Finally, it does not consider the presence of other ions such as Mg^{2+} , Ba^{2+} or the presence of CO_2 ; for

this is necessary to consider the concentrations of each ion present in the solution (Villafañe, 2010).

The (Eq. 1) represents the Langelier equation used for the calculus of the saturation pH. This equation involves the negative logarithm of the calcium concentration (pCa), the concentration of bicarbonate ($pHCO_3$) and the equilibrium constant (pK).

$$pH_s = pCa + pHCO_3 - pK \quad (Eq. 1)$$

This equation can be expressed in terms of the negative logarithm of the bicarbonate ionization constant (pK_2) and the negative logarithm of the solubility product constant (pK_{sp}), which is the (Eq. 2).

$$pH_s = (pK_2 - pK_{sp}) + pCa + pAlcalinity \quad (Eq. 2)$$

Therefore, the saturation index is the determinate by (Eq. 3).

$$IS = pH - pH_s \quad (Eq. 3)$$

This model will be used to determine in a quality way if there will be bicarbonate scale formation based on the value obtained of the LSI. According to the Langelier model, when the index value obtained is negative, there is no forming incrustation because the $CaCO_3$ is going to be dissolved by the water. When the index value obtained is positive the formation of the incrustation is going to happen, which means that the $CaCO_3$ is going to precipitate. Finally, when the value obtained is equal to zero this means that the system is in equilibrium.

4.1.2 STIFF-DAVIS SATURATION INDEX (ISSD)

The Stiff-Davis saturation index is an extension of the Langelier index to determine the formation of incrustations of CaCO_3 produced by the formation water of the oilfields. This model allows engineers to predict the tendency to deposit CaCO_3 in order to overcome limitations produced by the water with high solid content and its impact on the scale formation. Stiff Davis index is based on the saturation level concept, but the solubility product is modified empirically. This index calculation is recommended, especially for oilfields.

This model used three equations that are an extension of the Langelier model to obtain the saturation index.

$$ISSD = pH - pH_s \quad (Eq. 4)$$

This equation is the last used to determine the Stiff-Davis index (ISSD), which is the difference between the saturation pH calculated and the formation water's real pH. The saturation pH is calculated by the (Eq. 5), including the K constant of salinity which can be determined by using the correlation between the molar ionic strength and the temperature shown in Fig. 9, which is in the literature available for Stiff-Davis saturation index calculation, the composition and the temperature of formation water (Villafañe, 2010).

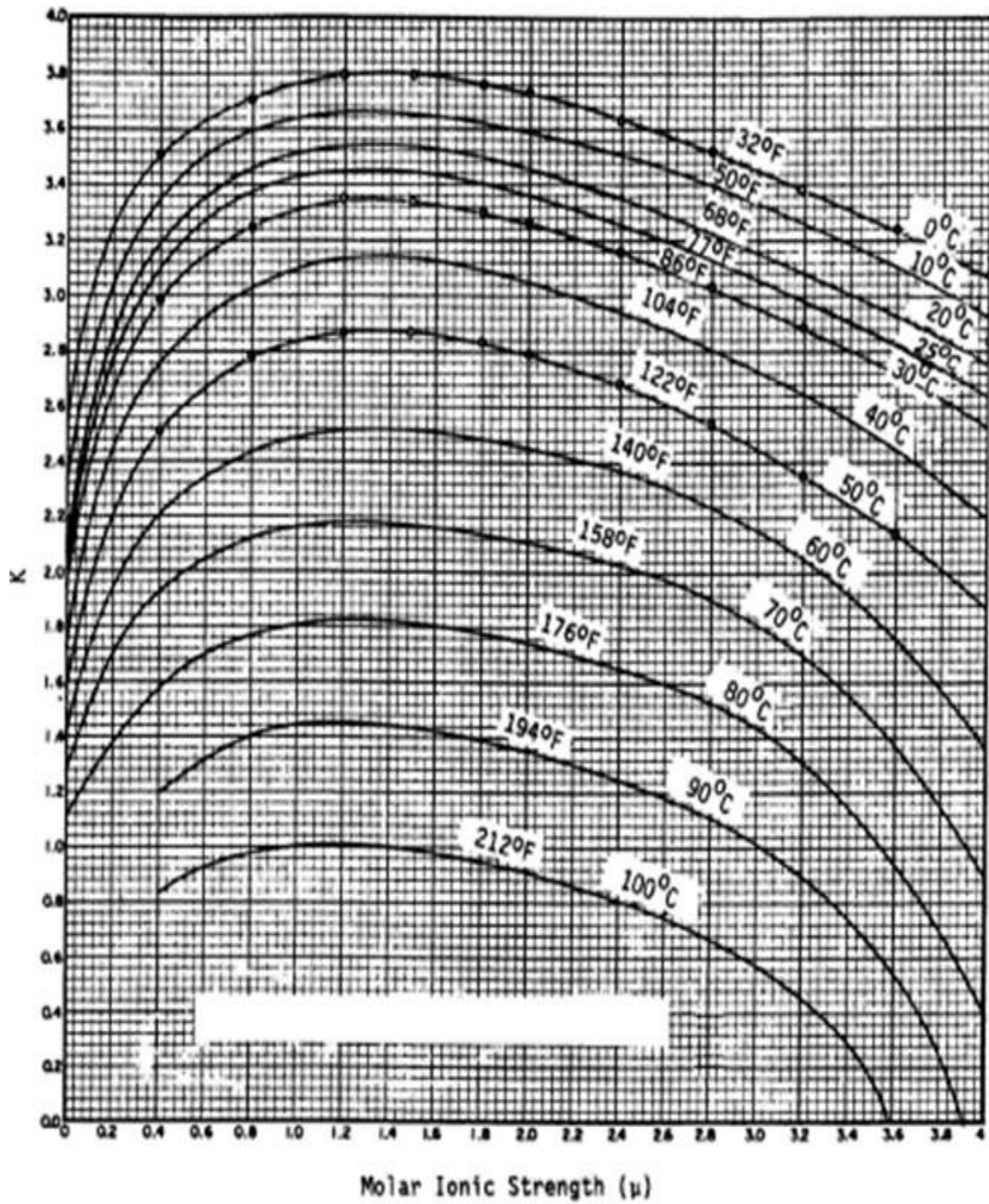


FIG. 9. CORRELATION FOR K VALUES DETERMINE (PATTON, 1995)

$$pH_s = K + pCa + pAlk \quad (Eq. 5)$$

Therefore, the Stiff-Davis equation can be expressed and used as the (Eq. 6).

$$ISSD = pH - K - pCa - pAlk \quad (Eq. 6)$$

The K value is determined by using the correlation (Fig. 9) between the ionic force and the temperature of formation water. For the calculus of the ionic force (Si) the (Eq. 7) is used in which the Ci is the concentration of ions in mole per liter and the Zi is the ion valence.

$$Si = \frac{1}{2} (C_1 Z_1^2 + C_2 Z_2^2 + \dots + C_n Z_n^2) \quad (Eq. 7)$$

The Stiff-Davis saturation index interpretation is the same as the Langelier saturation index, which will allow engineers to predict CaCO₃ incrustations.

4.1.3 ODDO-TOMSON SATURATION INDEX

The Oddo-Tomson index model is widely used in the Hydrocarbon Industry for scale formation determine, but in this study is not used because the program for this calculations which is the ScaleCorr 6.0 could not be used, because lack of license. It is described in the based on it importance in the Hydrocarbon Industry. This model includes corrections for the fugacity effects and the changes in CO₂ solubility, considered a function of the temperature, pressure and percentage of water with an additional improvement which is that the pH is not necessary. The restrictions that this model has been three. The first one is that de ionic force has to be between 0 to 4; the temperature has to be between 0 to 200°C, and finally, the pressure has to be between 0 to 20000 psig. The Oddo-Tomson index calculus is based on four equations for biphasic fluid and three equations for monophasic (Villafañe, 2010).

In the case of biphasic systems, the next four equations (Eq. 8), (Eq. 9), (Eq. 10) and (Eq. 11) are used to determine the formation of incrustations produced by CaCO_3 . These equations involved new terms such as the fugacity of CO_2 ($f_g^{\text{CO}_2}$), the CO_2 molar fraction in the gas phase ($y_g^{\text{CO}_2}$), the CO_2 molar fraction in the surface ($y_t^{\text{CO}_2}$), the ionic force (Si), the million standard cubic feet produced in standard conditions (MMscf), the barrels of water daily produced (BWPD) and the barrels of crude oil daily produced (BOPD).

In the case of monophasic systems two equations (Eq. 12) and (Eq. 13) are used for determine the saturation index and one new value is added, which is the CO_2 concentration in aqueous phase ($C_{aq}^{\text{CO}_2}$). And when the pH is measured, engineers can use the (Eq. 14) directly.

This model differs from the Langelier and Stiff-Davis models because they consider thermodynamic ideality, but Oddo-Tomson considers the deviations of the activity coefficient depending on temperature, pressure or ionic force changes. The error range of the other models produced by the measure of pH taken in the surface is eliminated because it is not necessary.

$$l_s = \log \left[\frac{(Ca^{+2})(HCO_3^-)^2}{py_g^{CO_2} f_g^{CO_2}} \right] + 5,85 + 15,19 \times 10^{-3} T - 1,64 \times 10^{-6} T^2 - 5,27 \times 10^{-5} p - 3,334(Si)^{0,5} + 1,431(Si) \quad (Eq. 8)$$

$$pH = \log \left[\frac{H_3O^+}{py_g^{CO_2} f_g^{CO_2}} \right] + 8,60 + 5,31 \times 10^{-3} T - 2,253 \times 10^{-6} T^2 - 2,237 \times 10^{-5} p - 0,990(Si)^{0,5} + 0,658(Si) \quad (Eq. 9)$$

$$f_g^{CO_2} = \exp \left[p \left(2,84 \times 10^{-4} - \frac{0,255}{T + 460} \right) \right] \quad (Eq. 10)$$

$$y_g^{CO_2} = \frac{y_t^{CO_2}}{1 + \frac{p f_g^{CO_2} (5 \text{ BWPD} + 10 \text{ BOPD}) \times 10^{-5}}{MMscf (T + 460)}} \quad (Eq. 11)$$

$$l_s = \log \left[\frac{(Ca^{+2})(HCO_3^-)^2}{C_{aq}^{CO_2}} \right] + 3,63 + 8,68 \times 10^{-3} T + 3,55 \times 10^{-6} T^2 - 6,56 \times 10^{-5} p - 3,42(Si)^{0,5} + 1,37(Si) \quad (Eq. 12)$$

$$pH = \log \left[\frac{H_3O^+}{C_{aq}^{CO_2}} \right] + 6,39 - 1,198 \times 10^{-3} T + 7,94 \times 10^{-6} T^2 - 3,53 \times 10^{-5} p - 1,067(Si)^{0,5} + 0,559(Si) \quad (Eq. 13)$$

$$l_s = \log \left[(Ca^{+2})(HCO_3^-)^2 \right] + pH - 2,76 + 9,88 \times 10^{-3} T + 0,61 \times 10^{-6} T^2 - 3,03 \times 10^{-5} p - 2,348(Si)^{0,5} + 0,77(Si) \quad (Eq. 14)$$

4.2 CO₂ CORROSION PREDICTION

CO₂ corrosion prediction is absolutely important in order to determine the impact of CO₂ present in the fluid flow of production in the facilities of production. The principal methods used are De Waard-Milliams and De Waard-Lotz, in which cases the operational parameters are considered (temperature and pressure) and allows engineers to know the corrosion rate. This means the wear of the production pipe per year due to corrosion.

4.2.1 DE WAARD-MILLIAMS METHOD

The De Waard and Milliams method is a predicting model for CO₂ corrosion based on the correlation between temperature and partial pressure of CO₂ in order to obtain the corrosion rate in millimeters per year, as it is shown in (Eq. 15).

$$\text{Log}(CR_t) = 7.96 - \frac{2320}{T + 273} - 5.55 \times 10^{-3}T + 0.67 \log(PCO_2) \quad (\text{Eq. 15})$$

This equation has changed after comparing it with the experimental results obtained, specifically the constants of the formula shown in (Eq. 16).

$$\text{Log}(CR_t) = 5.8 - \frac{1710}{T + 273} + 0.67 \log(PCO_2) \quad (\text{Eq. 16})$$

4.2.2 DE WAARD-LOTZ MODEL

To validate the De Waard-Milliams model, De Waard -Lotz generates a model of corrosion rate based on the velocity in the absence of surface scale with a parallel resistance model. This was plasm in the next equation, in which V_{cr} is the corrosion rate to determine, V_r is the independent flow contribution denoting the reaction rate, and V_m is the flow-dependent contribution denoting the mass transfer rate (Ossai, 2012).

$$\frac{1}{V_{cr}} = \frac{1}{V_r} + \frac{1}{V_m} \quad (Eq. 17)$$

To determine the value of V_r and V_m two more equations were modeled, the (Eq. 18) to determine V_r value and the (Eq. 19) to determine V_m , where U is the liquid flow rate and D_h is the hydraulic diameter of the pipe.

$$\text{Log}(V_r) = 4.93 - \frac{1119}{T + 273} + 0.58 \log(P_{CO_2}) \quad (Eq. 18)$$

$$V_m = 2.45 \frac{U^{0.8}}{D_h^{0.8}} P_{CO_2} \quad (Eq. 19)$$

According to the equations determined, this model considers the influence of the fluid in the possible corrosion process presented and, in the pipeline, directly. This model validates and gives a clearer vision of the corrosion's behavior and the effect on the pipeline.

4.3 MATERIAL SELECTION DIAGRAM

According to Jose Vera from DNV the material selection of a production pipeline can be determined by the partial pressure of CO_2 and partial pressure of H_2S , considering that according to the calculus the scale formation can be discarded (Vera, 2016).

This material selection is based on the effect of CO_2 and H_2S present in the fluid flow on the structure of the pipeline. This diagram is valid for high-performance steels subsoil-structures such as the casing tube and the production pipe. In the Fig. 10 different stainless steels are shown such as J55 which is casing tube material, 13-Cr which is a material of pipe production. As it is shown in the diagram the different CRA's (13-Cr MSS, several duplex steels and Ni-based) are specified with their respective boundaries of stress. In the left side are 13-Cr MSS and the duplex steels for low H_2S content and high CO_2 content, where the

corrosion impact will be done by CO₂ as Sacha's case. In the right side are the Ni-based alloys for more corrosive environments beginning in high H₂S content and low CO₂ content environments to high H₂S content and high CO₂ content environments.

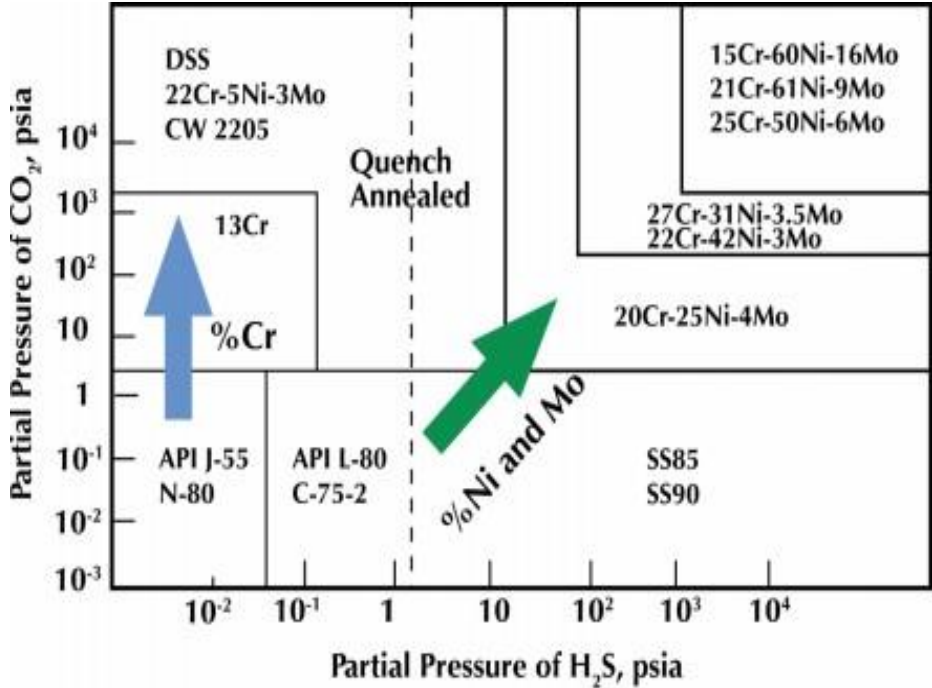


FIG. 10. MATERIAL SELECTION DIAGRAM (VERA, 2016)

CHAPTER 5: RESULTS AND DISCUSSION

5.1 HYDROCARBON INDUSTRY IN THE WORLD

Nowadays, most of the reserves are of heavy crude oil, so the technology and operational conditions have to change day by day. One of the principal ways to illustrate and confirm it is by studying the relation oil reserve-production. This rate gives a clear idea of each country's oil industry current situation and the comparison with its previous years lets engineers know if something happens. BP Company every year makes a statistical review of the world reserves, production and consumption of the different energy sources called "Statistical Review of World Energy". A simple graphic of the rate reserves-production such as Fig. 11 can help to analyze the actual hydrocarbon industry's situation.

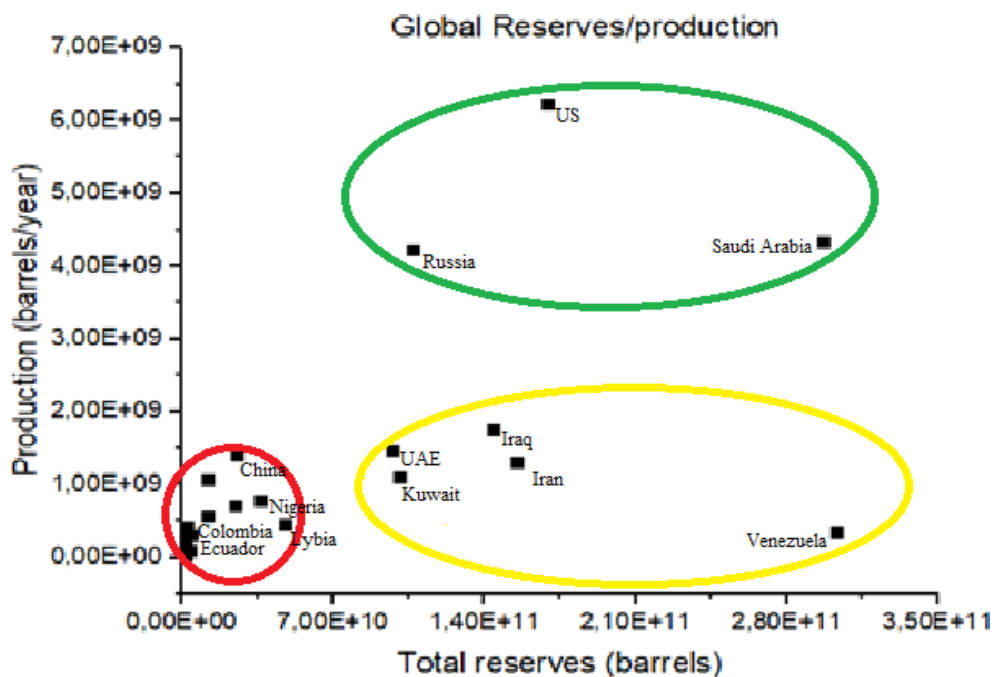


FIG. 11. RELATION RESERVE-PRODUCTION OF OIL COUNTRIES
(STATISTICAL REVIEW OF WORLD ENERGY BP, 2020)

According to the values showed in the Statistical review made by BP, there are three kinds of oil country based on their rate reserves-production. The first ones and the most important are those with big reserves and high production (green), such as Saudi Arabia, country with one of the highest oil production and one of the biggest proved oil reserve. Its current proved reserve is 297.6 thousand million barrels with a production of 11.83 million of barrels per day, and if this is compared to previous years, both factors have increased. Therefore, Saudi Arabia is an oil country with a good future projection of 68.9 years to take advantage of its reserve and which guarantees it that it can supply its necessities and remain in the market competitively.

The second ones are those countries with big reserves but with low production (yellow), such as Iran, Iraq or Venezuela. In these cases of Iran, Iraq and Venezuela, their rates reserves-production are too high, which means that they are not making the most of their resources because something affect their industries, in which cases are their geopolitics situations. This kind of example let engineers have a broader point of view and understand that they have to consider all of the possible factors that can affect the hydrocarbon industry.

The third ones are those with small reserves and low production (red), such as China ,Ecuador or Colombia. China represents a particular case which can be found in the small reserves and low production section with the particularity, that even when its reserves are small, it has a rate of reserves-production of is 18.7 years; this happens because its production is too low. By comparing Chinese oil production with its consumption, an important fact can be determined: its consumption is almost four times its production. Based on this previous information and checking the data of the last years, it can be inferred that China is one of the principal importers of crude oil globally and according to the historical; there is an increase of 9.5% to the previous years.

According to the reserves' actual situation, production, and consumption of oil countries, the world's reserves continue decreasing, and day-by-day new technologies are necessary to increase the reserves founded and increase the production. Thus, it is necessary to have controlled all of the factors that can affect the hydrocarbon industry, such as political situation, environmental conditions, types of reserves, methods for recovery, or even international conflicts.

5.2 AMERICA'S OIL INDUSTRY SITUATION

The Americas situation is particular because there are countries for each section. Here engineers have particular cases to study to understand how different factors in a country can influence in its industries but especially in hydrocarbon industry, which is the principal source of energy for most countries. For this is necessary to locate in America's situation by the information supplied by important companies such as BP Company in its annual review "Statistical Review of World Energy". Information that is showed in Table. 5 and Table. 6.

TABLE. 5. RESERVE, ANNUAL PRODUCTION AND RATE OF PRINCIPAL OIL COUNTRIES IN AMERICA
(STATISTICAL REVIEW OF WORLD ENERGY BP, 2020)

Country	Reserve (Thousand million Barrels)	Anual Production (Thousand million barrels per year)	Rate (years)
Argentina	2,4	0,23	10,61
Brazil	12,7	1,05	12,09
Colombia	2	0,32	6,18
Ecuador	1,6	0,19	8,26
Peru	0,9	0,05	17,36
Trinidad&Tobago	0,2	0,03	6,68
Venezuela	303,8	0,34	906,68
Otros	0,5	0,04	11,61
Total	324,1	2,25	144,04

TABLE. 6. OIL CONSUMPTION OF PRINCIPAL OIL COUNTRIES IN AMERICA
(STATISTICAL REVIEW OF WORLD ENERGY BP, 2020)

Country	Consumption (MBPD)	Production (MBPD)	Percentage of consumption
US	19400	17045	113,82
Canada	2403	5651	42,52
Brazil	2398	2877	83,35
Mexico	1733	1918	90,35
Venezuela	356	918	38,78
Colombia	347	886	39,16
Argentina	599	620	96,61
Ecuador	249	531	46,89
Peru	262	142	184,51
Trinidad&Tobago	39	82	47,56
Others	1294	118	1096,61

United States is one of those particular cases. US is a country with moderate reserves and has the highest production globally as it is shown in Table. 5 and Table. 6, which is attributed to its technology of production, which is fracking. This technology consists of a high-pressure injection of water, sand, and chemicals to break the rock and release the hydrocarbon. However, its production is not enough because its consumption is bigger than its production as it is shown in Table. 6 with a percentage of consumption around 114%. US even has a strategic oil reserve of 645 million barrels, but its rate reserve-production is only 11 years, which does not guarantee their future oil market.

Venezuela's case is important to analyze because it has the biggest reserve in the world with 303,08 thousand million barrels and has the right technology to take advantage of its resources and lead the market, but its political situation does not allow it to be like that. Every year its production decrease and its economic situation have been affected and the

stability of its population. If the government does not take any action, Venezuela's hydrocarbon industry situation will go from bad to worse.

As we can see in the Table. 5 the America situation is absolutely alarming, specially for those countries in which cases their economy depends on the hydrocarbon industry such as Ecuador. In those cases is necessary to find an alternative energy source in the next 5 to 8 years because, according to their rate reserve-production, those are the years that their reserves have left if they do not find more proved reserves.

5.3 ECUADORIAN HYDROCARBON INDUSTRY

5.3.1 PRINCIPAL PROCESSES FOR PRODUCTION IN ECUADOR

5.3.1.1 SEISMIC SURVEY

A seismic survey is the first step for determining where an oilfield is, its capacity and potential. A seismic survey is necessary to begin oil exploitation; it consists of generating seismic waves and measuring the time it takes for the wave to reflect and be detected by the geophone (Villamarin and Carrera, 2010).

5.3.1.2 DRILLING

In Ecuador, the cluster drilling method is used to reduce the number of platforms and deforestation (Villamarin and Carrera, 2010).

5.3.1.3 HYDROCARBON EXTRACTION

The fluid is obtained by the lateral or lower aquifers' hydraulic thrust or by the different artificial methods such as electro submersible pumping, mechanic pumping, gas lift or by water reinjection method currently used.

Once at the surface level, the fluid from the different wells is integrated through a manifold system. Therefore, the fluid is directed to the process plants to separate the oil, gas and the formation water (Villamarin&Carrera, 2010).

5.3.2 ACTUAL SITUATION OF ECUADORIAN OILFIELDS

According to the information acquired from the Statistical Review of World Energy 2020 and to the Ecuadorian Hydrocarbon Minister, Ecuador's oil production is around 531 thousand barrels per day, and a total volume of 1.6 thousand million barrels of crude oil proved reserve. This means that Ecuador has a reserve-production rate of 8.4 years if the production is constant and they do not find new reserves.

The scheme for the exploitation and production of crude oil in Ecuador is doing by the public industry called Petroecuador EP and the private industries such as Repsol, Petroriental, Andes Petroleum and other private companies. The public company Petroecuador EP manages approximately 1.39 thousand million barrels of the total proved reserve of Ecuador, and the private companies manage the other 242 million barrels of the proved reserve (Chiluisa and Merino, 2018).

According to the Ecuadorian hydrocarbon secretariat, Ecuador has 71 production blocks between Petroamazonas EP and the private companies (hydrocarbon secretariat, 2016).

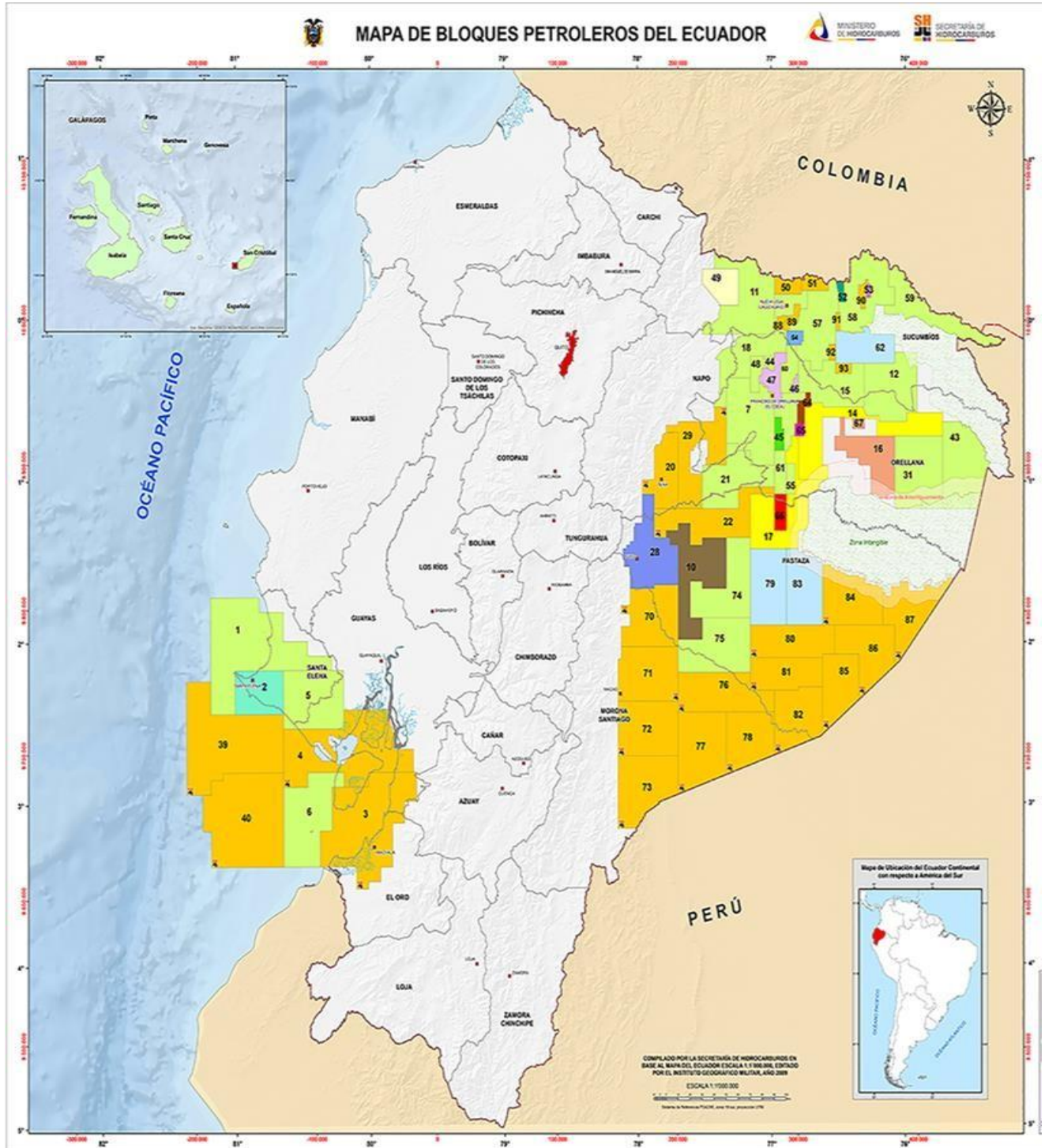


FIG. 12. ECUADORIAN OILFIELDS (HYDROCARBON SECRETARIAT, 2016)

In the annual report of Ecuador's hydrocarbon potential, the detail of the percentage of accumulate production, proved reserves, probable reserves, and possible reserves of each oilfield are described. Most of the oilfields in Ecuador are mature oilfields, in which cases there are no future projects to recover the probable and possible reserves, but oilfields such as Sacha or Auca represent the future oil sources. This happens because according to The Minister of Non-Renewable Energy Sacha, Shushufindi and Auca have the 21%, 19% and 18% respectively of the proved reserves of Ecuador, representing 806.2 million barrels (Chiluisa and Merino, 2018).

*TABLE. 7. PROBABLE AND POSSIBLE RESERVES PRINCIPAL OILFIELDS
(ANNUAL REPORT OF THE HYDROCARBON POTENTIAL OF ECUADOR, 2018)*

Oilfields	Probable reserves	Percentage represented	Possible reserves	Percentage represented
Sacha	56,98	22	70,18	9,6
Auca	41,44	16	92,84	12,7
ITT	54,39	21	496,35	67,9

The three main oilfields that lead the proved reserves in Ecuador are Sacha, Shushufindi and Auca; the ones for the total probable reserves are Sacha, ITT and Auca with 22%, 21% and 16% respectively; and the ones for the total possible reserves are ITT, Auca and Sacha with 67.9%, 12.7% and 9.6% respectively as it is shown in Table. 7. This is why these four oilfields are important for the hydrocarbon industry and the country's economy's stability; they represent the present and the future of the industry.

5.4 SCALE FORMATION RESULTS

5.4.1 LANGELIER RESULTS

According to the results obtained in the Langelier index calculus for the range of temperature used (120°F-160°F), there will not be any carbonate formation. The ideal range of the Langelier index is between -0.3 and 0.3. This allows engineers to discard a possible formation of carbonate scale.

TABLE. 8. LANGELIER METHOD RESULTS FOR SACHA'S FORMATION WATER

Temperature (°F)	pHs	LSI
127	6,37	-0,04
135	6,29	0,04
140	6,24	0,09
145	6,2	0,13
150	6,15	0,18
160	6,06	0,27
165	6,01	0,32
170	5,97	0,36
176	5,91	0,42

For this calculation was necessary to take the pH of 6,3 of the wellhead as a constant and the TDS, calcium concentration and carbonate concentration as constant with the wellhead values shown in the Table. 8, those values taken from the Table. 4 were applied in the in the Lenntech Calculator for each temperature for Langelier Saturation Index calculation.

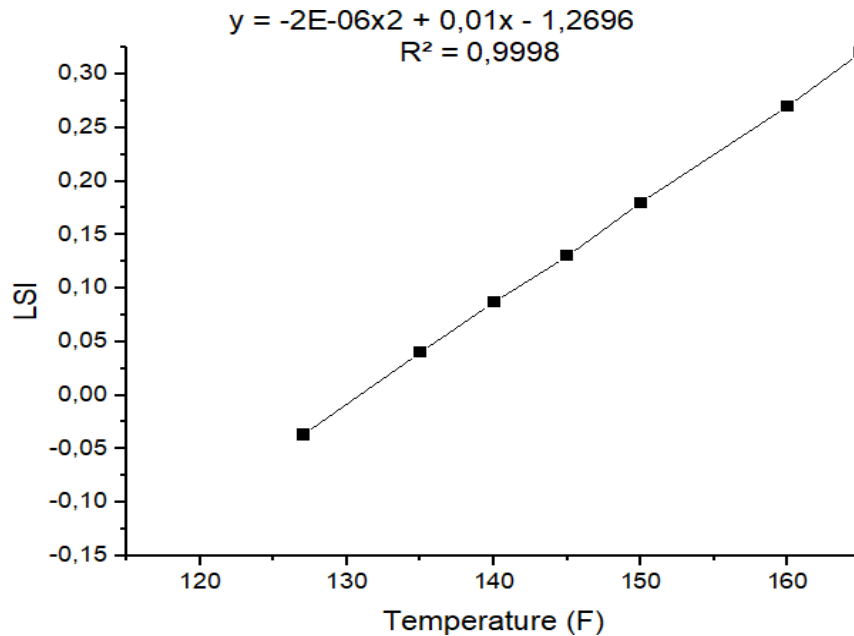


FIG. 13. GRAPHIC TEMPERATURE VS LSI

In the Table. 8 and Fig. 13 is shown the behavior of the salts dissolved in the formation water, especially the total solid dissolved, the ion calcium and the bicarbonate based on the temperature changes and by taking the pH of the sample of formation water as a constant. This method is not exact because it does not cover the total temperature from the downhole to the wellhead of the oilfield; but it clarifies what can happen. The results obtained show the difference between the real pH and the saturation pH, the Langelier saturation index that varies from -0.11 to 0.27. These values indicate that there will not be scale formation, but it is necessary to consider that the real pH can vary with the temperature changes, representing a possibility of scale problems. In this case, an inhibitor can be used to prevent and cover that possibility.

5.4.2 STIFF-DAVIS RESULTS

For the Stiff-Davis index calculation was necessary to convert the ions' concentrations taken from the Table. 4 to molar concentration (Table. 9) in order to calculate the molar ionic strength (Table. 10) with the values obtained and by applying them in the Eq. 7.

TABLE. 9. RESULTS OF IONS' MOLAR CONCENTRATIONS OF THE FORMATION WATER

Ions	Concentration (mg/L)	Concentration (mol/L)
Na ⁺	5759	0,2504
Ca ²⁺	720	0,018
Mg ²⁺	96	0,004
HCO ₃ ⁻	434	0,0071
Fe ²⁺	4,5	0,00008
SO ₄ ⁻	3	0,00003
Cl ⁻	10200	0,2877

TABLE. 10. RESULT OF MOLAR IONIC STRENGTH

Si (ionic force)
0,316775

By using the Fig. 9, the K value can be determined. First we have to put in the X axis in the value of the molar ionic strength obtained and go up to the temperature of the point of the studied conditions, which in this case is the wellhead temperature and at this point the value of the Y axis is the value of K factor.

TABLE. 11. K VALUE OBTAINED BY THE CORRELATION DIAGRAM

K Factor
2,32

TABLE. 12. STIFF-DAVIS SATURATION INDEX OBTAINED

Temperature	pH	K factor	pCa	pALK	ISSD
127	6,33	2,32	1,75	2,15	0,11

Finally the ISSD value is obtained by the application of the Eq. 6 in which the pCa and pALK are the negative logarithm of calcium concentration and alkalinity respectively and the K value obtained before shown Table 11. The ISSD value obtained is in the ideal range, and there is not a big difference from the Langelier value obtained for the Wellhead conditions. This indicates that there will not be a scale formation problem. For a larger study of the oilfield is necessary to know the characteristics and conditions not only from the wellhead but also downhole. This is important to understand the behavior in all of the production pipeline to determine if it is necessary to reduce the possibility of incrustations formation.

According to the values obtained with the two methods of scale formation and based on the history of corrosion problems in Sacha, the incrustation problems can be discarded. This allows us to analyze the possibility of corrosion by CO₂, which will be analyzed by De waard-Milliams and De waard-Lotz methods.

5.5 CO₂ CORROSION PREDICTION RESULTS

5.5.1 DE WAARD-MILLIAMS RESULTS

For De waard-Milliams model is necessary to determine the partial pressure of CO₂ for each section of the production pipeline with the respect temperature of each point. For this, it was necessary to determine a pressure profile (Table. 13) and temperature profile which is the second column in Table. 14 by using the wellhead conditions (Table. 4) and the

downhole conditions (Table. 3) and divide them by a referential depth of Hollin formation which according to Romero & Gomez is 10160 fts or 3100 meters. This Value obtained is going to be the pressure or temperature gradient, which is going to be multiplied with the meters of the point that we want to determine and added to the head pressure taken from Minda Theses in this case or head temperature taken from Table. 4. Then, the partial pressure was obtained by the product between the concentration which according to Leiva is 9% and the operational pressure as it is shown in the Table. 13.

TABLE. 13. PARTIAL PRESSURE OF CO₂

Depth (m)	Pressure (Psia)	Concentration CO ₂ (%)	Partial Pressure CO ₂ (Psia)	Partial Pressure CO ₂ (mpa)
0	340	9	31	0,21
300	738	9	66	0,46
600	1136	9	102	0,71
900	1534	9	138	0,95
1200	1933	9	174	1,20
1500	2331	9	210	1,45
1800	2729	9	246	1,69
2100	3127	9	281	1,94
2400	3525	9	317	2,19
2700	3923	9	353	2,43
3000	4322	9	389	2,68
3100	4454	9	401	2,76

By using the De waard-Milliams equations (Eq. 15 and Eq. 16) the corrosion rate is determined. It is important to consider the two equations in order to obtain a bigger range of corrosion rate. This allows making a better material selection because it shows the

possible sceneries and the impact that CO₂ will have on the production pipeline. The corrosion's behavior presented in the different sections is shown in the Table. 14 and the corrosion rate profile obtained for each equation is shown in the Fig. 14 and Fig. 15.

TABLE. 14. DE WAARD-MILLIAM RESULTS OF CORROSION RATE PROFILE

Partial Pressure CO ₂ (mpa)	Temperature (°C)	Log(crt) (Eq. 15)	CRT (Eq. 15) (mm/yr)	Log (crt) (Eq. 16)	CRT (Eq. 16) (mm/yr)
0,21	52,77	0,33	2,15	0,10	1,25
0,46	58,04	0,67	4,64	0,41	2,55
0,71	63,32	0,90	7,89	0,61	4,11
0,95	68,59	1,09	12,18	0,78	6,02
1,20	73,87	1,25	17,81	0,92	8,38
1,45	79,14	1,40	25,12	1,05	11,26
1,69	84,42	1,54	34,50	1,17	14,76
1,94	89,69	1,67	46,41	1,28	18,97
2,19	94,97	1,79	61,36	1,38	24,02
2,43	100,24	1,90	79,96	1,48	30,02
2,68	105,52	2,01	102,89	1,57	37,10
2,76	107,28	2,05	111,62	1,60	39,73

These values shown in Table. 14 were obtained by using the partial pressure profile and temperature profile in the Eq. 15 and Eq. 16 described for De waard-Milliams model, all those calculations were made in Excel. This gave us estimated corrosion rates for each point in the production well the operacional conditions and the CO₂ content.

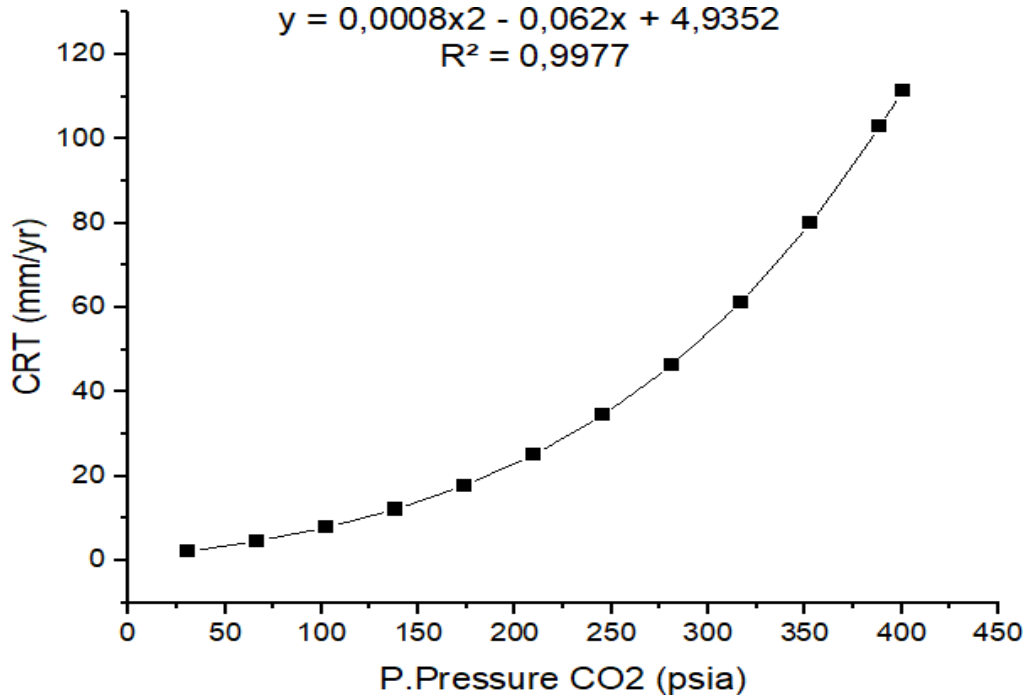


FIG. 14. CORROSION RATE PROFILE OBTAINED BY (EQ. 15)

According to the Fig. 14, the corrosion rate increased directly proportional to the pressure and temperature. This equation indicates a corrosion rate range from 2,15 mm/yr to 111,33 mm/yr. These values are alarming because by using a carbon steel pipe with the typical dimensions used in Sacha's oilfield which has a thickness of 0,237 inches (6,02 millimeters); this represents that it will be necessary to change the pipeline in the best case once every three years or in the worst case eighteen times every year or less by considering the use of coating.

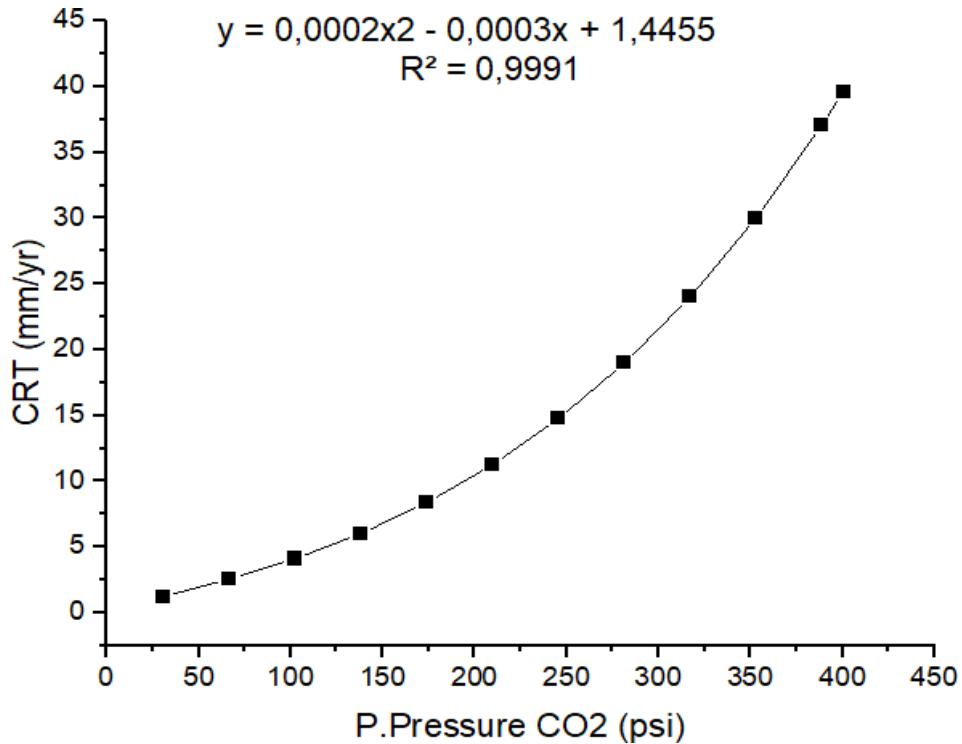


FIG. 15. CORROSION RATE PROFILE OBTAINED BY (EQ. 16)

According to the Fig. 15 the corrosion rate increased directly proportional to the pressure and temperature, this equation indicates a corrosion rate range from 1,25 mm/yr to 39,64 mm/yr. These values obtained are reasonable, even if we consider using a carbon steel pipe with coating by reducing the changes produced to once every six years and in the worst-case six times or less every year. In the (Eq. 16), the temperature and partial pressure are not the only parameters considered for determining the corrosion rate, liquid flow rate is considered. This parameter is necessary to be considered because it affects and increase the corrosion problems presented in production pipeline.

Based on the values obtained with both equations (Eq. 15) and (Eq. 16), the CO₂ corrosion is a significant problem to be considered.

5.5.2 DE WAARD-LOTZ RESULTS

For the correct application of this model, it was necessary to take the characteristics of the actual production pipeline used in Sacha's oilfield. The diameter used was 4,5 inches, which in meters is 0,1143 meters. Another factor considered is the liquid flow rate obtained by the continuity equation and the value is 2,83 meters per second. This model covers more aspects of the corrosion by CO₂ considering a significant fluid dynamic aspect.

TABLE. 15. LIQUID FLOW RATE AND DIAMETER FOR A TYPICAL SACHA'S OILFIELD

Liquid flow (BFPD)	Conversion factor (m ³ /s)	Diameter (m)	Dh (m)	Area (m ²)	Liquid flow rate (m/s)
15728,94	1,84x10 ⁻⁶	0,1143	0,1143	0,01026	2,83

In order to perform the De waard-Lotz model it was necessary to use the temperature and partial pressure profiles calculated for the De waard-Milliams method, the liquid flow rate in meters per second and the hydraulic diameter of the pipeline, as it is shown in the Table. 15. For the corrosion rate calculation is necessary to calculate the flow independent contribution by applying the Eq. 18 denoting the reaction rate and the flow-dependent contribution by applying the Eq. 19 denoting the mass transfer rate. Finally with both values the Eq. 17 was applied in order to determine the corrosion rate.

TABLE. 16. CORROSION RATE OBTAINED BY DE WAARD-LOTZ MODEL

Depth (m)	Partial Pressure CO ₂ (mpa)	Temperature (°C)	log (V _r)	V _r (mm/yr)	V _m (mm/yr)	V _{cr} (mm/yr)
0	0,21	52,77	1,10	12,68	6,74	4,40
300	0,46	58,04	1,35	22,55	14,62	8,87
600	0,71	63,32	1,51	32,72	22,51	13,34
900	0,95	68,59	1,64	43,84	30,40	17,95
1200	1,20	73,87	1,75	56,20	38,29	22,77
1500	1,45	79,14	1,85	70,03	46,18	27,83
1800	1,69	84,42	1,93	85,49	54,06	33,12
2100	1,94	89,69	2,01	102,74	61,95	38,65
2400	2,19	94,97	2,09	121,94	69,84	44,41
2700	2,43	100,24	2,16	143,25	77,73	50,39
3000	2,68	105,52	2,22	166,81	85,62	56,58
3100	2,76	107,22	2,24	174,91	88,16	58,62

According to the values obtained in the Table. 16 the CO₂ corrosion effect is significant and matches with the values obtained by the De waard-Milliams model shown in the Table. 14.

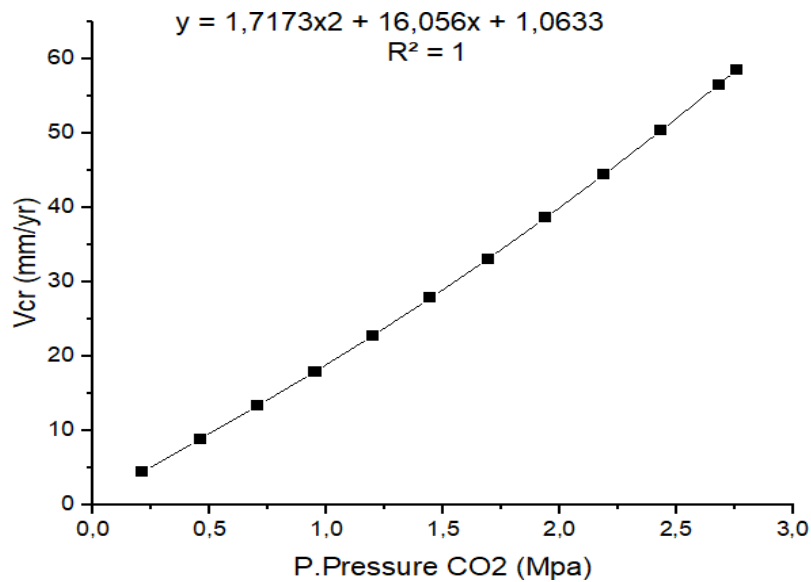


FIG. 16. CORROSION RATE PROFILE OBTAINED BY DE WAARD-LOTZ MODEL

As shown in the Fig. 16, the tendency indicates that the corrosion rate increases with all of the parameters involved in the model's equations.

5.6 DISCUSSION

As we see, Sacha is one of the main oilfields in Ecuador with a lifetime longer than the other ones based on the probable and possible reserves, which generates the necessity of a long lifetime facility. The use of new methods of recovery such as water reinjection increases the corrosion problems presented in oilfields. This recovery method is used in Sacha oilfield. This represented an important point to be analyzed to discard all of the possible corrosion problems and determine if it is necessary to use high-performance steel in the subsurface-surface section.

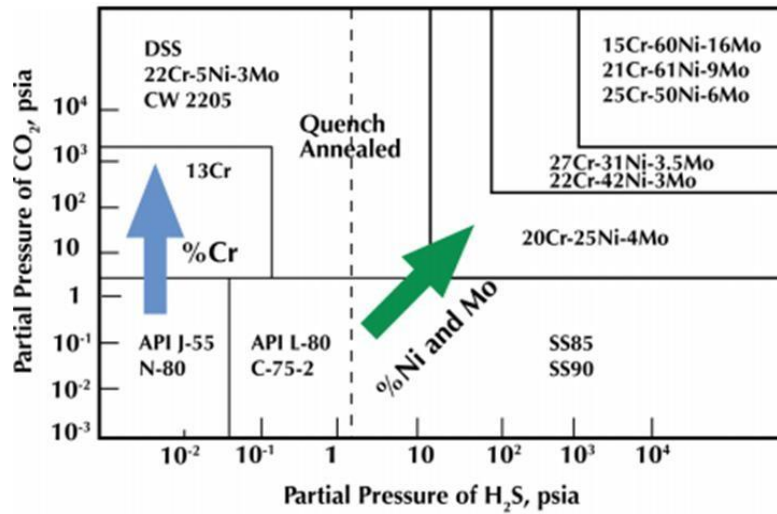
The data obtained by Langelier and Stiff-Davis models indicate that there will not be scale formation based on the Sacha's formation water characteristics. According to the Langelier model's values for the profile temperature from the wellhead or zero meters to 1800 meters, the index saturation is in the ideal range from -0,3 to 0,3. This value of index saturation gives the certainty that there will not exist scale formation and it was verified with an alternative method, the Stiff-Davis model.

The Stiff-Davis model considers the ions' concentrations; therefore, this model only applies to a specific point where the formation water was analyzed. The Stiff-Davis saturation index calculation for the wellhead gives us a value of 0,11 which means that there will not exist incrustation formation. By comparing the values obtained by the Langelier and Stiff-Davis method, we can see that they are similar, and both are in the ideal range of saturation index to eliminate the possibility of scale formation.

For the CO₂ corrosion problems, the De waard-Milliams and De waard-lotz methods are the principal used. According to De waar-Milliams results obtained, the CO₂ corrosion is significant, it increases with the temperature and pressure, making the corrosion rate increase to 111,33 millimeters per year according to the (Eq. 15) and to 39, 64 millimeters per year according to the (Eq. 16). Both graphics obtained from the results have a similar tendency, are described by a second-degree equation and give a clear idea of the impact generated in the pipeline especially in the downhole section.

In the case of De waard-Lotz model other parameters are considered especially the influence of the flow of fluid related to the specifications of the pipe used. In this case, the maximum flow rate was considered and the specifications of a typical pipe used in Sacha's oilfields; By using these values and the pressure and temperature profile, the range of corrosion rate obtained goes from 4,399 millimeter per year in the wellhead to 58,61 millimeter per year in the downhole. By comparing these values obtained with the De waard-Milliams results of the corrected equation (Eq. 16) it is clear that the flow of the fluid has an influence in the corrosion produced by CO₂.

These results obtained by the two models allow engineers to know that they can not use steel carbon pipe, because, with these corrosion rates, it will be necessary to change the structure constantly. Some engineers could think that it can be controlled by using a correct chemical treatment, but with these high rates of corrosion, it could not be.



According to the Material Selection Diagram (Fig. 10), using the partial pressure of CO_2 which is in the order of 10^2 and the partial pressure of H_2S which is in the order of 10^{-1} , and by considering the operational variables (temperature and pressure) and the actual Sacha's reserves conditions, the correct material to be used in Sacha's oilfields is 13-Chrome to reduce the maintenance cost and to avoid possible operational problems, which can represent a future disaster. This high-performance steel compared with the API 5L GRB material (surface material recommended) used in Sacha's wells will increase the relation benefit-cost, especially based on thinking Sacha as an actual and future project for economy sustainable.

CONCLUSION AND RECOMMENDATIONS

The natural decline of reservoirs increases the water production, which significantly impacted reliability and operational continuity; therefore, the corrosion problems in all of the oilfield processes to the final product have increased. This is why oil industries have had to resort to new recovery methods such as water injection. This affects the operational conditions and changes the fluid of production characteristics such as chloride content, corrosive agents, water and sediments. For this reason it is necessary to do a correct analysis of the possibility of scale formation and the CO₂ or H₂S corrosion in order to take the correct material selection and the correct preventive method. Sacha is one of the most important reserves of crude oil of Ecuador for this moment and for the future based on its proved, probable and possible reserves and the country's profits. By the analysis of its actual characteristics, the principal operational problem that Sacha can present is CO₂ corrosion, and according to the values obtained, even if inhibitors are used, this will represent money for future changes; Therefore, in this case, a correct material selection can help to avoid or reduce the future problems and expenses. This is why a correct material selection in the hydrocarbon industry represents one of the most important processes. It is necessary to consider and study all of the variables when we are going to determine and select the corrosion processes to be studied. For future projects it is absolutely important to repeat the study with the other hydrocarbon industry processes such as transport or refining processes in order to determine the adequate material based on the actual conditions and optimize running costs.

REFERENCES

- ❖ Baby, P.; Rivadeneira, M.; Barragan, R.; Christophoul, F. Thick-Skinned tectonics in the Oriente Foreland basin of Ecuador. *Geo. Art.* [Online] **2013**. <https://sp.lyellcollection.org/content/377/1/59>
- ❖ BP. Statistical Review of World Energy 2020. *Chem.Rev.* [Online] **2020**. <https://www.bp.com/en/global/corporate/energy-economics/statistical-review-of-world-energy/oil.html> (accessed July 5, 2020)
- ❖ Cantos, J. Diseño de una Planta para el Tratamiento de Agua de Formación para su Reinyección en el Campo Sacha. Theses [Online], Escuela Politecnica Nacional, UIO, January 2017. <https://bibdigital.epn.edu.ec/handle/15000/16987> (accessed July 3, 2020)
- ❖ Carbotecnia S.A de C.V. La tendencia incrustante, corrosiva o equilibrada de un agua, y el índice de saturación de langelier. <https://www.carbotecnia.info/aprendizaje/quimica-del-agua/como-se-calcula-el-indice-de-langelier/> (accessed August 5, 2020)
- ❖ Castellan W. *Fisicoquímica*, 2nd ed.; Massachussts, 1987; appendix 12.
- ❖ Chiluisa, C.; Merino, J. Informe anual del potencial hidrocarburífero del Ecuador. *Chem. Rev.* [Online] **2018**, Vol 2.2. <https://www.recursoyenergia.gob.ec/wp-content/uploads/2019/11/Informe-Anual-del-Potencial-Hidrocarburi%CC%81fero-del-Ecuador-2018.pdf> (accessed July 10, 2020)

- ❖ Fernandez, K.; Gaibor, N. Redimensionamiento de los Sistemas de Producción de Fluidos y de Reinyección de Aguas de Formación, en las Estaciones Sacha Norte 1, Sacha Norte 2 y Sacha Sur, para que Cumplan los Estandares de producción del Bloque 15. Theses [Online], Escuela Politecnica Nacional, UIO, July 2009. <https://bibdigital.epn.edu.ec/handle/15000/1607> (accessed July 5, 2020)

- ❖ Himipex Oil Group. Metal protection from the hydrogen sulfide corrosion. <http://himipex.com/the-problem-of-protection-of-metal-from-the-hydrogen-sulfide-corrosion> (accessed July 18, 2020)

- ❖ IEAGHG. Corrosion and Material Selection in CCS Systems. *Chem. Rep.* [Online] **2010**. <https://www.globalccsinstitute.com/resources/publications-reports-research/corrosion-and-materials-selection-in-ccs-systems/> (accessed July 25, 2020)

- ❖ Introduction to CRA tubulars: Metallurgy and Material Selection for Corrosive Environments. In Houston, TX, Sep 25, 2017; CRA Tubular and Well Integrity Technical Symposium: Houston, TX.

- ❖ Kemmer, F. *The Nalco Water Handbook*, 2nd ed.; McGraw-Hill: New York, 1988.

- ❖ Landazuri, S. Diagnostico y Optimización del Sistema de Reinyección de Agua de Formación en el Campo Sacha. Theses [Online], Universidad Central del Ecuador, UIO, May 2015. <http://www.dspace.uce.edu.ec/handle/25000/7702> (accessed July 5, 2020)

- ❖ Lenntech USA LLC. <https://www.lenntech.com/calculators/langelier/index/langelier.htm> (accessed August 30, 2020)

- ❖ Leiva, L. Estudio de factibilidad para la Utilización del Gas Asociado que se Produce en Tres Plataformas del Campo Sacha para Generación Electrica. Theses [Online], Universidad Central del Ecuador, UIO, February 2018.
<http://www.dspace.uce.edu.ec/handle/25000/15635> (accessed July 5, 2020)

- ❖ Merino, L. Estudio Tecnico Economico para la Perforación Horizontal de Pozos de Producción de Petroleo en la Arena “U” del Campo Sacha Sur. Theses [Online], Escuela Politecnica Nacional, UIO, March 2013.
<https://bibdigital.epn.edu.ec/handle/15000/6017> (accessed July 6, 2020)

- ❖ Minda, V. Evaluación de los Diferentes Indices de Saturación de Carbonatos para Determinar la Tendencia Incrustante o Corrosiva de los Fluidos de Producción en el Bloque 16 de la Amazonía Ecuatoriana. Theses [Online], Universidad Tecnologica Equinoccial, UIO, February 2016.
<http://repositorio.ute.edu.ec/handle/123456789/16803> (accessed July 8, 2020)

- ❖ Nausha, B.; Qystein, M.; Jackson, J.; Jenkins, A.; Mélot, D.; Scheie, J.; Vittonato, J. Corrosion - The Longest War. *Chem. Rev.* [Online] **2016**, Vol 28, 34-49.
<https://www.slb.com/-/media/files/oilfield-review/03-corrosion-english> (accessed July 13, 2020)

- ❖ Ortega, D. Modelado de Métodos Predictivos para la Formación de Incrustaciones en Tuberías con Flujo de Agua de Producción. Theses [Online], Universidad Central de Venezuela, CCS, November 2017. <http://saber.ucv.ve/handle/123456789/17179> (accessed July 20, 2020)

- ❖ Ossai, C. Predictive modelling of Wellhead Corrosion due to Operating conditions: A field data approach. *Chem. Art.* **2012**, 1-8. doi:10.5402/2012/237025
- ❖ Patton, C. *Applied Water Technology*, 2nd ed .; Campbell Petroleum Series: Oklahoma, 1995.
- ❖ Petroamazonas. La Producción del Campo Sacha Supera los Setenta Mil Barriles Diarios de Petróleo. <https://www.petroamazonas.gob.ec/?p=10166> (accessed August 2, 2020)
- ❖ Phull, B.; Byrd, J.; Castillo, R.; Laury, D.; Campbell, J.; Pasini, N.; Brady, S. *Manual del Estudiante Corrosión Básica*, Vol. 2.01.; NACE Internacional, US, 2014.
- ❖ Pierre, R. *Handbook of Corrosion Engineering*, 2nd ed.; McGraw-Hill: New York, 2012.
- ❖ Prange, F. Mechanical properties and corrosion resistance of oil Well Tubing. *Corrosion.* **1959**, 15(2), 13-18. doi:10.5006/0010-9312-15.2.13
- ❖ Romero, J.; Gomez, F. Estudio de los Efectos de la Inyección de Agua en los Yacimientos “U” y “T” de la Formación Napo del Campo Sacha. Theses [Online], Escuela Politecnica Nacional, UIO, April 2010. <https://bibdigital.epn.edu.ec/handle/15000/1973> (accessed July 8, 2020)
- ❖ Secretaría de Hidrocarburos. Mapa de Bloques Petroleros. <http://www.historico.secretariahidrocarburos.gob.ec/mapa-de-bloques-petroleros/> (accessed July 18, 2020)

- ❖ Stiff, H. A., & Davis, L. E. A method for predicting the tendency of oil field waters to deposit calcium carbonate. *Journal of Petroleum Technology*. **1952**, 4(09), 213-216. doi:10.2118/952213-g
- ❖ Vera, J. Managing Internal Corrosion of Subsea Pipelines and Risers. In Houston, TX, Oct 15-18, 2018; DNV GL offshore pipeline day Congress: Houston, TX, 2018.
- ❖ Villafaña, C. Estudio de la Aguas de Formación y Producción: Gestión del Riesgo por la Problemática de Incrustaciones. Theses [Online], Universidad Central de Venezuela, CCS, November 2017. <http://saber.ucv.ve/handle/123456789/17214> (accessed July 22, 2020)
- ❖ Villamarin, O.; Carrera, A. Diseño de una Plataforma Típica de Producción Petrolera en la Amazonía Ecuatoriana. Theses [Online], Escuela Politecnica Nacional, UIO, June 2010. <https://bibdigital.epn.edu.ec/handle/15000/2230> (accessed July 8, 2020)
- ❖ Vilorio, D. *Uso de inhibidores en la industria de los hidrocarburos*. Caracas, 2014; pp 21-24.
- ❖ Zurita, L. Estudio de la Velocidad de Corrosión en una Tubería de Producción de Petróleo con Relación a la Dosificación de un Inhibidor Químico. Theses [Online], Escuela Politecnica Nacional, UIO, January 2017. <https://bibdigital.epn.edu.ec/handle/15000/17032> (accessed July 10, 2020)

ANNEX

Annex A: Ecuadorian Hydrocarbon Industry Annual Budget (2019)



PETROAMAZONAS EP
EJECUCION INVERSIONES - REPORTES - INFORMACION CONSOLIDADA
EJECUCION DEL PRESUPUESTO
 Expresados en Dolares
 Programa + 27
 Actividad - Grupo Gasto - Item - FTE -
 DEL 1RO DE ENERO AL 31 DE DICIEMBRE DE 2019

EJERCICIO 2019

	DESCRIPCION	ASIGNADO	DEVENGADO	SALDO POR DEVENGAR	% EJE
BLOQUE CENTRO					
11200	Inversiones en Consorcios y Subsidiarias	164,947,874	301,192,121	63,755,753	61%
11300	Sismica y geofisica	0	0	0	0%
11400	Exploración	0	0	0	0%
11400	Perforación - Desarrollo	1,061,806	104,069	537,818	47%
11500	Facilidades de producción	33,904,015	38,673,698	-4,769,683	114%
14000	Asset Fijo (cuenta solo en Orade)	2,899,666	2,429,779	469,887	84%
TOTAL BLOQUE CENTRO		202,813,442	142,799,667	60,013,774	70%
BLOQUE OESTE					
11200	Inversiones en Consorcios y Subsidiarias	674,193,260	666,663,565	7,529,695	99%
11300	Sismica y geofisica	0	0	0	0%
11400	Exploración	0	0	0	0%
11400	Perforación - Desarrollo	2,569,902	-787,673	3,357,575	-31%
11500	Facilidades de producción	3,614,615	2,152,091	1,462,524	60%
14000	Asset Fijo (cuenta solo en Orade)	152,307	121,209	30,899	80%
TOTAL BLOQUE OESTE		680,529,585	668,149,192	12,380,393	98%
BLOQUE ESTE					
11300	Sismica y geofisica	0	0	0	0%
11400	Exploración	0	0	0	0%
11400	Perforación - Desarrollo	177,740,890	155,125,046	22,615,845	87%
11500	Facilidades de producción	92,165,197	87,263,449	5,001,748	94%
14000	Asset Fijo (cuenta solo en Orade)	420,503	8,049	412,454	2%
TOTAL BLOQUE ESTE		270,329,590	242,216,543	28,113,047	90%
BLOQUE NORTE					
11200	Inversiones en Consorcios y Subsidiarias	459,964,306	447,893,314	11,970,991	97%
11300	Sismica y geofisica	0	0	0	0%
11400	Exploración	0	-1,044	1,044	0%
11400	Perforación - Desarrollo	37,091,582	30,870,124	6,221,458	83%
11500	Facilidades de producción	33,328,870	32,567,657	761,213	98%
14000	Asset Fijo (cuenta solo en Orade)	1,436,856	1,119,008	317,848	78%
TOTAL BLOQUE NORTE		531,721,213	512,447,059	19,274,154	98%
OGE SECTORIAL					
11300	Sismica y geofisica	0	0	0	0%
11400	Exploración	0	0	0	0%
11400	Perforación - Desarrollo	0	0	0	0%
11500	Facilidades de producción	26,330,610	24,056,187	2,274,423	91%
14000	Asset Fijo (cuenta solo en Orade)	0	-302,400	302,400	0%
TOTAL OGE SECTORIAL		26,333,610	23,753,787	2,579,823	90%
EXPLORACION					
11300	Sismica y geofisica	165,484,153	-1,676,945	167,161,097	-1%
11400	Exploración	0	0	0	0%
TOTAL EXPLORACION		165,484,153	-1,676,945	167,161,097	-1%
TOTAL EJECUCION INVERSIONES PETROAMAZONAS		1,877,211,593	1,587,689,304	289,522,288	85%
		0	1,587,689,304	0	

PETROAMAZONAS EP
EJECUCION DE COSTOS Y GASTOS E INVERSIONES - REPORTES - INFORMACION CONSOLIDADA
EJECUCION DEL PRESUPUESTO

Expresados en Dólares

Programa = 27

Actividad - Grupo Gasto - ítem - FTE -

DEL 1RO DE ENERO AL 31 DE DICIEMBRE DE 2019

EJERCICIO 2019

	DESCRIPCION	ASIGNADO	DEVENGADO	SALDO POR DEVENGAR	% EJEC
GASTOS DE PERSONAL					
10	Remuneraciones Mensuales	144,940,711	147,469,093	-2,528,383	102%
12	Aportes y Cargas Sociales	45,182,478	45,150,197	32,281	100%
14	Indemnizaciones	16,857,431	14,096,771	2,760,659	84%
TOTAL GASTOS DE PERSONAL		206,982,619	206,716,062	266,557	100%
SERVICIOS GENERALES					
20	Válidos y Gastos de Viaje	417,072	208,024	209,049	50%
21	Servicios Médicos	640,830	413,038	226,791	64%
24	Seguros	8,187,373	8,886,614	300,759	97%
25	Publicidad y Propaganda	176,000	120,818	55,182	69%
26	Arrendamientos	170,835,925	166,926,053	3,909,873	98%
27	Capacitación	100,000	57,640	42,360	58%
TOTAL SERVICIOS GENERALES		181,360,200	176,612,187	4,748,013	97%
SERVICIOS OPERACIÓN U MANTENIMIENTO					
28	Alquiler y Transporte	15,460,353	15,102,154	358,199	97%
28	Servicios de Operación	417,366,922	404,132,739	13,234,183	97%
29	Servicios de Mantenimiento	69,269,392	68,469,158	800,233	99%
TOTAL SERVICIOS OPERACIÓN Y MANTENIMIENTO		502,128,867	487,704,051	14,424,815	97%
MATERIALES Y SUMINISTROS					
40	Materiales de Operación	98,937,925	85,794,583	13,143,342	87%
41	Materiales de Mantenimiento	19,230,099	18,889,258	340,842	98%
42	Combustibles y Lubricantes	158,775,358	158,238,460	536,899	100%
TOTAL MATERIALES Y SUMINISTROS		276,943,983	262,922,300	14,021,682	95%
IMPUESTOS Y TRANSFERENCIAS					
60	Impuestos y Contribuciones	23,913,477	23,371,785	541,692	98%
TOTAL IMPUESTOS Y TRANSFERENCIAS		23,913,477	23,371,785	541,692	98%
GASTOS FINANCIEROS					
70	Gastos Financieros	29,151,379	29,123,150	28,229	100%
TOTAL GASTOS FINANCIEROS		29,151,379	29,123,150	28,229	100%
TOTAL COSTOS & GASTOS SIN DD&A		1,220,480,525	1,186,449,535	34,030,989	97%
DEPRECIACIÓN y AMORTIZACIÓN					
90	Gasto de Depreciación y amortización	0	2,030,149,055	-2,030,149,055	0%
TOTAL DEPRECIACION Y AMORTIZACION		0	2,030,149,055	-2,030,149,055	0%
TOTAL EJECUCION PETROAMAZONAS		3,097,692,117	2,774,138,839	323,553,278	90%